

Development & Formulation of Taxifolin Loaded Niosomes for its Anticancer Effect on MCF-7 Breast Cancer

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ABSTRACT

In order to increase Taxifolin's solubility and bioavailability, this study set out to create and assess niosomes of Taxifolin powder by the thin film hydration approach. A nonionic surfactant called span 60 and a membrane stabiliser called cholesterol were utilised. Particle size distribution, zeta potential, and light scattering analysis were all examined. Niosomes demonstrated a mean particle size between 106.5 ± 7.42 nm and 115.8 ± 11.61 nm, with particle size increasing as the surfactant and cholesterol molar ratios were raised. The percentage yield was found to be in range $44 \pm 0.235\%$ to $81 \pm 0.216\%$. The optimized batch (FC6) had zeta potential of -27.7 ± 2.37 mV and 92.78 ± 3.14 % entrapment efficiency. Taxifolin loaded niosomes demonstrated sustained release of Taxifolin. The anticancer potential of compound was evaluated using an *in vitro* cytotoxicity assay on MCF-7 cell lines. Finally, associated to free Taxifolin, cytotoxic impact of Taxifolin loaded niosomes on MCF-7 cell line was found to be substantially improved by the MTT assay. Result demonstrated that solubility and bioavailability of naturally derived Taxifolin phytochemicals can be increase by loading it into niosomes, thus enhance their anticancer efficacy.

Keywords: Taxifolin, Phytochemical, Niosomes, MCF-7 Breast Cancer Cell.

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INTRODUCTION

The greater safety and biocompatibility of bioactive substances produced from natural sources make them the preferred choice over manufactured medications.¹ Herbal remedies made from plants have a long history of use as a remedy. Natural medicines, like Taxol have recently gained popularity as an alternative therapy and have received clinical approval for management of cancer patients.^{2,3} Plumbagin, a yellow dye that is officially generated from naphthoquinone, is another example. Multiple cancer cell lines have shown that plumbagin induces cell death and cell cycle arrest, including melanoma, breast, and lung cancers.⁴ Phytochemicals have a limited therapeutic window, are poorly soluble in water, have a low bioavailability, penetrate cells poorly, have a hepatic disposition, and are quickly absorbed by normal tissues, all of which reduce their effectiveness in fighting cancer. In order to increase selectivity, efficacy, safety, dose reduction, and patient compliance, drug delivery structures based on nanocarriers have been established for a variety of natural bioactives.⁵ Various chemicals produced from edible plants have been associated with cancer treatment and chemoprevention.^{6,7} Among these organic substances naturally occurring polyphenolic chemicals, including flavones, isoflavones, flavanones, flavonols, chalcones, and anthocyanins, are rich in flavonoids, a family of phytochemical agents. The potential of dietary flavonoids to prevent and treat cancer has been thoroughly investigated in both laboratory and

clinical research.⁸ Flavonoids have been demonstrated to have a variety of pharmacological actions and have a chemical structure with quercetin. The compound taxifolin, formally known as dihydroquercetin, has potent antioxidant properties.^{9,10} Comparable to quercetin, which has been shown to trigger cell death, Taxifolin shares many of the same characteristics. Current medical practice makes extensive use of taxifolin for its antioxidant, anticancer, and anti-virus capabilities, among others. This compound has also found widespread application in the food and health care industries.¹¹ Taxifolin have low solubility made it difficult to absorb and metabolize in the body, resulting in a considerable reduction of its bioavailability and effectiveness.¹² The dissolving rate and drug solubility were found to be correlated with precise surface area of materials, by smaller drug particles having a larger effective area contacting the media.¹³ To increase Taxifolin's solubility and bioavailability, it is essential to use nanotechnology to produce uniformly small niosomes as nanocarrier. Drug delivery involves use of several nanocarriers, with niosomes being one of the most effective. A vesicular nanoparticle known as a niosome is made up of helper lipids like cholesterol and non-ionic surfactants that can make the bilayer stiffer.¹⁴ Niosomes possess a great deal of potential to specifically target tumor cells and exhibit a regulated release within them. Due to their remarkable attributes such as Uncomplicated fabrication, affordability, excellent stability, and the capability to encase both

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Table 1: Optimization on the basis of Surfactant and Cholesterol.

Sr. No.	FC	Taxifolin: Span 60: Cholesterol; (% w/w/w)	% Practical Yield	Average Particle Size (nm)	Zeta Potential (mV)	% EE	PDI
1.	F1	1:1:1	44 ± 0.235%	106.5 ± 7.42	-23.8 ± 1.32	53.51 ± 3.31	0.094 ± 0.083
2.	F2	1:1:2	57 ± 0.169	108.7 ± 11.84	-17.7 ± 1.74	64.51 ± 3.31	0.103 ± 0.12
3.	F3	1:2:1	62 ± 0.261	105.2 ± 8.78	-18.4 ± 1.32	73.32 ± 3.73	0.213 ± 0.086
4.	F4	1:2:2	69 ± 0.198	100.6 ± 10.57	-25 ± 2.74	76.68 ± 2.87	0.420 ± 0.08
5.	F5	1:3:3	75 ± 0.283	112.1 ± 9.42	-20.7 ± 1.46	81.34 ± 3.27	0.226 ± 0.089
6.	F6	1:4:4	81 ± 0.216%	115.8 ± 11.61	-27.7 ± 2.37	92.78 ± 3.14	0.191 ± 0.98

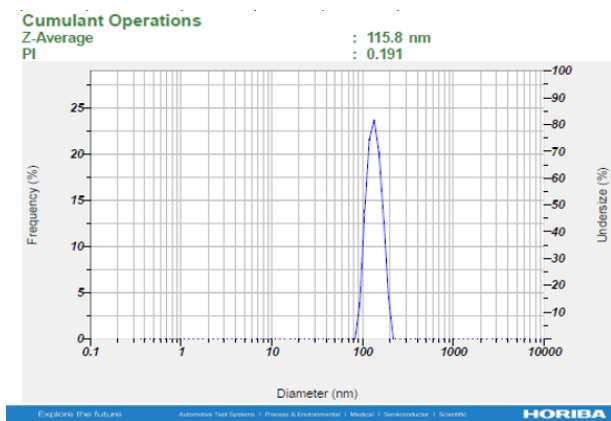


Figure 1: Particle size of FC6 formulation.

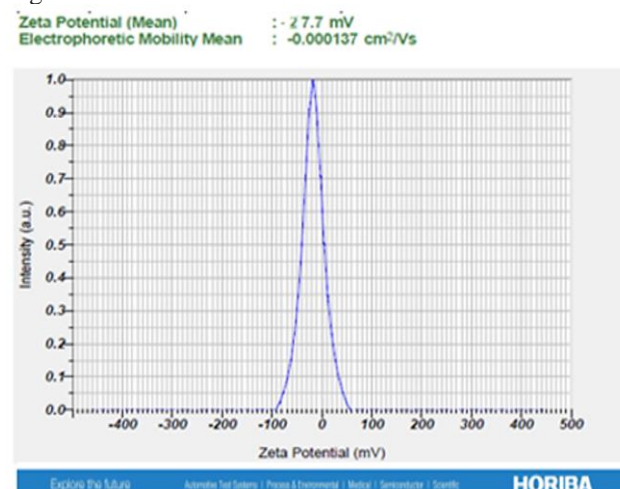


Figure 2: Zeta potential of FC 6 formulation.

hydrophobic and hydrophilic drugs niosomes are considered a viable alternative to other delivery systems, especially liposomes.^{15,16}

MATERIAL AND METHOD

Material

Taxifolin powder was obtained from Yucca Enterprises Mumbai, Cholesterol and Span 60 was obtained from Loba Chemie. All other materials were of analytical grade. The MCF-7 cell lines were bought from NCCS, Pune, India. MTT reagent, Dimethylsulfoxide, were procured from Sigma-Aldrich Chemical Private Ltd India.

Method of preparation

Thin film hydration method was used to create Taxifolin-encapsulated niosomes, as previously reported. In round-

bottom flask with chloroform, exact molar ratios of Taxifolin, span 60 & cholesterol (1:4:4), and DCP were dissolved. Following that, a thin film was created and the chloroform was evaporated by rotary evaporator set to 60°C and 100 rpm for 30 minutes (Labline PBV 7D). The resulting lipid film was hydrated for one hour at 60°C using 10 milliliters of pH 7.4 phosphate buffer in order to produce niosomes formulations. To produce niosomes with a consistent size distribution, the resultant milky solution was collected and consequently sonicated on ultrasonic bath for 30 min at 50°C.¹⁷ This suspension is then allowed to freeze overnight at -20 °C refrigerator. To obtain powder, the frozen niosomes was then subjected to lyophilize. The formed niosomes was used for evaluated.^{18,19}

Characterization of niosomes

Determination of percentage yield

The percentage yield can be used to calculate how much product is produced once the process is finished. The percentage yield of all batches of formulations was calculated by dividing measured weight of formulations with theoretical weight and multiplying with 100 by subsequent equation.²⁰

$$\text{Percentage Yield} = (\text{Practical Yield} / \text{Theoretical Yield}) \times 100$$

Analysis of PDI, Particle size, and zeta potential

PDI, particle size, and zeta potential were measured, which employs dynamic light scattering technology. Prior to analysis, the formulations were diluted and transferred into a cuvette, where assessment were conducted at 25°C ± 5°C. Zeta potential was assessed using a specialized cuvette with an applied potential of 3.3 mV. For determining electrophoretic mobility, the niosomes formulations were placed in a zeta potential cuvette.^{21,22}

% Entrapment efficiency

Entrapment efficiency represents the proportion of the active component encapsulated within the niosomes. Entrapment efficiency of niosomal formulation was evaluated using a centrifugation technique. Niosomes were isolated through ultracentrifugation at 18,000 rpm for 25 minutes, and entrapment efficiency was assessed by computing difference between total drug captured inside niosomes and unencapsulated drug available in supernatant. Quantity of free active constituent in supernatant was assessed spectrometrically at a wavelength of 288 nm. Entrapment efficiency (EE) % was then derived using expression below.²³

$$\% \text{ Entrapment Efficiency} = (\text{Total drug} - \text{Free drug in supernatant} / \text{Total drug}) \times 100$$

Table 2: Model, equation, and regressions of Taxifolin release from niosomes.

Model	Equation	R ²
Zero-Order	$y = 3.2208x + 30.224$	R ² = 0.662
First Order	$y = 0.0459x + 1.1984$	R ² = 0.3168
Higuchi	$y = 19.397x + 9.6295$	R ² = 0.9005
Kors-peppas	$y = 0.933x + 0.9459$	R ² = 0.5341

DSC analysis

DSC was utilized to evaluate the thermal behavior and physical form of pure Taxifolin, the physical mixture, and the optimized formulation. The analysis was performed under nitrogen flow within a temperature 25°C to 260°C, by heating increment of 10°C per minute, on samples weighing 3-5 mg that were placed in aluminium pans.^{24,25}

In vitro drug release

An *in vitro* drug release experiment was executed to investigate release profiles of free Taxifolin and Taxifolin-loaded niosomes. Dialysis bag method was employed for this dissolution study. Separate dialysis bags, each

containing either free Taxifolin or lyophilized Taxifolin-loaded optimized formulation were soaked overnight in phosphate buffer (pH 7.4). After sealing both ends of bags, they were connected to paddles of a dissolution test instrument (Electrolab Dissolution Tester TDL-08L, India) containing phosphate buffer (pH 7.4) as dissolution fluid. Wholemethod was upheld at $37 \pm 0.5^\circ\text{C}$ by a paddle rotational frequency of 100 rpm. At predefined time recesses, samples were taken out and exchanged by an equal volume of newly prepared medium to preserve saturated state. Quantity of drug released from formulation was measured using a UV spectrometrically at 288 nm.^{26,27}

Kinetic release study

The *in vitro* drug release data was evaluated using many kinetic models. Following are the models used to determine the niosome-release kinetics of Taxifolin: the Zero-order, First-order, Higuchi, and Korsmeyer-Peppas versions (KP). Determination coefficient (r^2) for each model was calculated to identify the optimal model for release data.^{28,29}

In vitro cytotoxicity (MTT assay) for anticancer activity

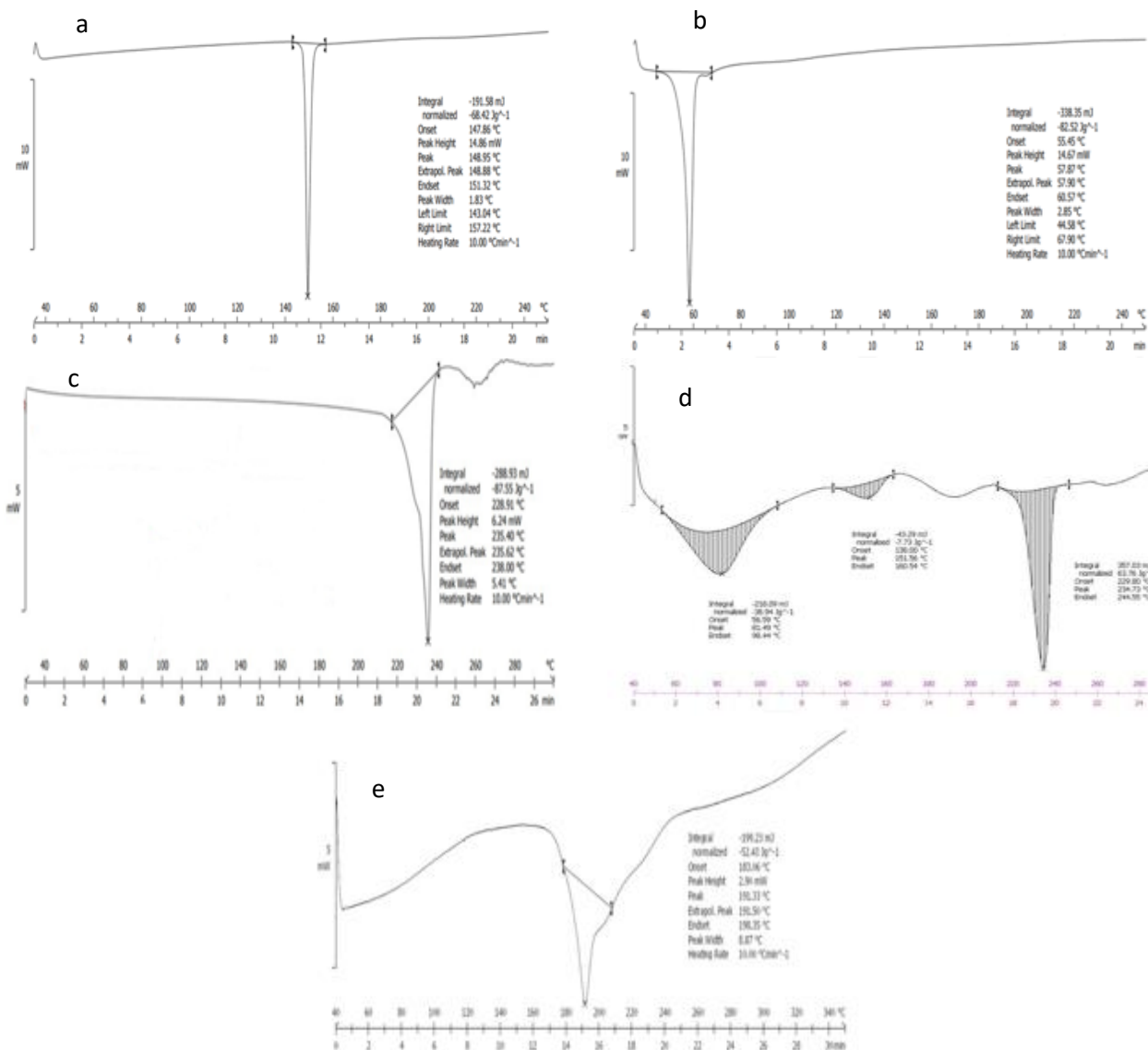


Figure 3: DSC thermogram of a. cholesterol, b. span 60, c. Taxifolin, d. physical mixture, e. Taxifolin niosomes formulation (FC6).

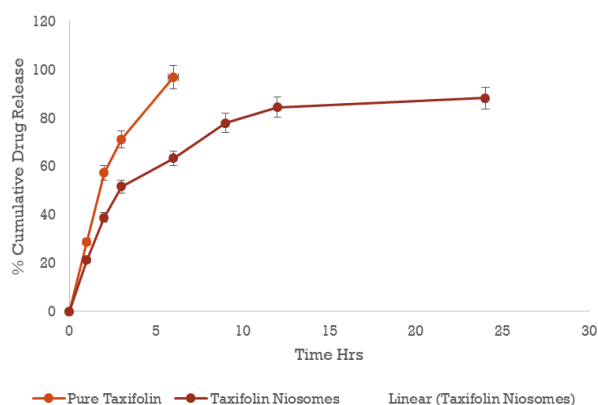


Figure 4: Taxifolin and Taxifolin niosomes *In vitro* release profiles in PBS 7.4 at 37 °C.

Preparation of cells

Non-essential amino acids and 10% foetal bovine serum were added to the MEM in which MCF-7 cells were cultured. The cells were sustained at 37°C in a 5% CO₂ environment, with entire medium being altered each 2 to 3 days.

MTT assay procedure

The cytotoxic effects of the niosomes formulation were assessed using the MTT assay with MCF-7. The MCF-7 cell line, sourced from NCCS, Pune, India, was grown in 96-well plates and humidified incubator with 5% CO₂ for 24 hours, achieving a near-confluent monolayer. Culture medium was then detached, and cells were subjected to seven concentrations levels of Taxifolin and Taxifolin-loaded niosomes (10, 25, 50, 100, 250, 500, and 1000 µg/ml). Cisplatin was used as a reference standard. Cells cultured without any test substances served as the control. Following a 24-hours incubation and 5% CO₂, cells were inspected under an inverted microscope for any morphological changes or signs of cell death.³⁰ Following this observation, 100 µl of freshly made medium and 10 microlitres of MTT dye were used to replace the original media.³¹ After that, the plates needed to be incubated for four hours at 37°C with 5% CO₂. Afterwards, 100 µl of a solubilisation solution was added to each well, and plates were thereafter left at 37°C with 5% CO₂ for an additional hour.³² The absorbance of formazan crystals at 570 nm was measured using a microplate reader to assess the cell survival percentage.^{33,34} Next, the percentage was calculated utilising formulas that have been published.^{35,36} Percent Cell Survival (%) = (Absorbance of Test / Absorbance of Control) * 100

RESULTS AND DISCUSSION

Determination of practical yield

% yield of FC6 formulation was highest as compared to other formulations. The percentage yield was found to be in range 44 ± 0.235% to 81 ± 0.216%.

Analysis of size of particle, PDI and zeta potential

Optimization of niosomes formulations was achieved by adjusting surfactant to cholesterol molar ratio to maximize entrapment efficiency, target size of particle, zeta potential,

and particle size distribution. Span 60 to cholesterol ratio was refined to identify optimal formulation. Particle size measurements, conducted using dynamic light scattering with a Horiba SZ 100, ranged from 106.5 ± 7.42 nm to 115.8 ± 11.61 nm. An increase in particle size was noted with higher concentrations of cholesterol and Span 60, likely due to the development of a more robust bilayer membrane. Using the PDI, we were able to determine how homogeneous the formulation was; values below 0.1 indicated very high homogeneity, while values above 0.3 indicated very high heterogeneity. The optimized formula (FC6) showed a PDI of 0.191±0.98, which means it was stable and homogeneous enough. The zeta potential was used to assess the stability of niosomes formulations. Typically, a zeta potential close to ±30 mV ensures stability for longer duration. The optimized batch exhibited a zeta potential of -27.7±2.37 mv. High negative zeta potential values generally reflect stable formulations and minimize aggregation (Figure 1 and Figure 2). The inclusion of Taxifolin in niosomes resulted in lower zeta potential values, contributing to increased formulation stability.³⁷

Determination of % entrapment efficiency

The entrapment efficiency data across various formulation codes (FC1 to FC6) indicate that FC6 stands out as the optimized formulation, achieving the highest entrapment efficiency. The optimized FC6 has an entrapment efficiency of 92.78% with a margin of error of 3.14%. The incorporation of hydrophobic medicines into niosome bilayers through electrostatic interactions has been demonstrated in numerous studies. Exceptional entrapment of Taxifolin in niosomes is due to its hydrophobic properties, which enhance its affinity for and incorporation into the bilayers. This high drug loading is advantageous for systemic administration, ensuring a substantial release of taxifolin at the locus of action. In comparison, other study reported a loading efficiency of 69% for lawsone-encapsulated niosomes with a particle size of 0.3microns (Table 1). The Taxifolin-loaded niosomes developed in this study not only demonstrate superior loading efficiency but also exhibit a reduced particle size relative to the sizes as reported earlier.^{38,39}

DSC analysis

The DSC thermogram technique was used to confirm the physical state of optimized niosomes formulation. DSC thermogram of cholesterol, span 60, Taxifolin, its physical mixture and optimized batch Taxifolin niosomes (FC6) was recorded and displayed in figure 3a, 3b, 3c, 3d, and 3e respectively. DSC analysis of pure Taxifolin, a prominent endothermic peak at 235.40°C was seen, demonstrating its purity. In the physical mixture unique endothermic peaks of the individual components were evident. DSC analysis of the optimized batch (FC6) was found to be 198.35 °C. It indicates that Taxifolin is dispersed in molecular level to excipients or Taxifolin is in amorphous form. It was also observed that the thermogram showed no additional peaks of available free Taxifolin, indicating that there had not been any precipitation and that the Taxifolin had been completely encapsulated in the vesicles.⁴⁰

In vitro release study

Diffusion bag release studies were executed in a phosphate buffer with a pH of 7.4 at 37 °C to determine *in vitro* release of both free Taxifolin and niosomes loaded with Taxifolin. Figure 4 shows that there is a substantial alteration in cumulative % release between free Taxifolin and Taxifolin niosomes formulation. Free Taxifolin was able to completely release its drug content within 6 hours when it was allowed to freely diffuse in solution form. Nevertheless, the release of Taxifolin from niosomes showed a biphasic pattern, beginning by a burst release of 21.24% during the first hour and continuing with sustained release for up to 24 hours. The release rate of Taxifolin from niosomes was much lower than that of free Taxifolin, remaining at $88.18 \pm 0.012\%$ for the entire 24 hours. Niosome bilayer loading was facilitated by Taxifolin's hydrophobic properties. Since there is no shell to allow for prolonged release, the release of free Taxifolin was extremely rapid; nearly all of the drug was liberated within 6 hours. Our formulation's Taxifolin release rate was much lower than that of free Taxifolin. These results demonstrated that niosomes can release Taxifolin in a controlled and sustainable manner. A key component of effective cancer treatment, according to research, is the prolonged release of medicines from niosomes. The release kinetics of paclitaxel-encapsulated niosomes were studied earlier. Following 17 hours, their formulation had a 95.6% drug release rate. Drug packed niosomes exhibit a persistent

release pattern, as demonstrated by the aforementioned research investigations.⁴¹

Kinetic release study

By comparing *in vitro* release data with several release kinetics models, drug release mechanism from niosomes was anticipated. Kinetic studies showed that R^2 Value of Higuchi Model was highest than that of others, i.e. ($R^2 > 0.9005$) near to one suggesting the Higuchi Kinetic Model was the best fit release model followed (Table 2). It indicates drug release from insoluble matrix.⁴²

In vitro cytotoxicity (MTT assay) for anticancer activity

The release that occurs in a controlled laboratory setting by employing the MTT assay, the researchers examined the anticancer effects of Taxifolin-loaded niosomes *in vitro* and contrasted them with free Taxifolin in solution, setting of human breast cancer MCF-7 cells, taxifolin niosomes and conventional cisplatin are examined. Because niosomes are more soluble and interact better with malignant cells, loading hydrophobic medicines into them can increase the toxicity of the resulting formulations. The results shown in figure 5 and figure 6 indicate that taxifolin-loaded niosomes have better anticancer activity than free taxifolin in solution and standard cisplatin. It is clear that the toxicity of the medicine increased when the amounts of Taxifolin were raised. The loading of Taxifolin into niosomes resulted in a notable increase in cytotoxicity. Taxifolin niosomes had a

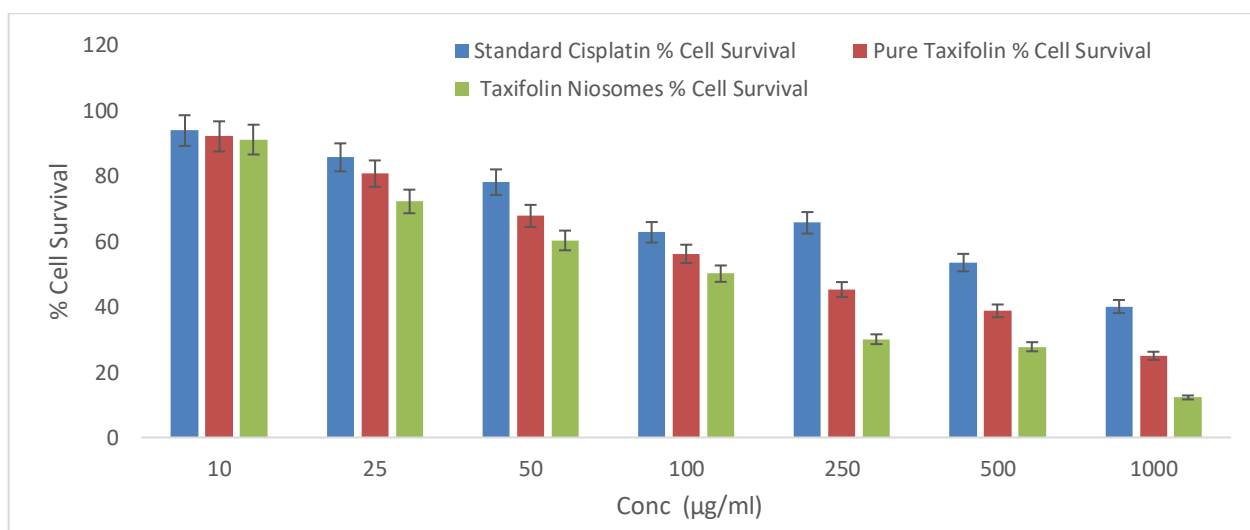


Figure 5: MTT cytotoxicity assay of cisplatin, pure Taxifolin, and Taxifolin niosomes (FC6) was determined at various concentrations ($\mu\text{g/ml}$) in MCF7 cells.

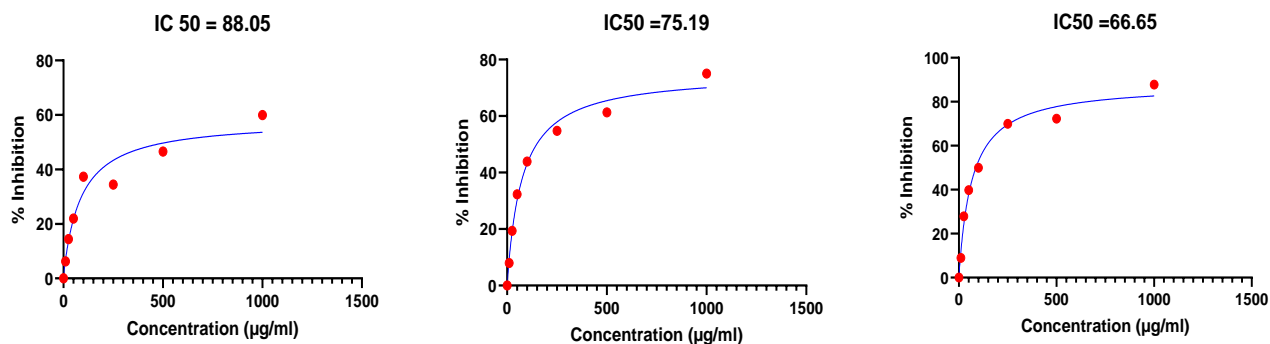


Figure 6: The bar graph indicating % cell survival of standard cisplatin, free Taxifolin and Taxifolin niosomes formulation against MCF-7 cell line.

viability of 12.23% at a concentration of 1000 µg/ml, compared to 24.95% for free Taxifolin. A drug's IC₅₀ value is concentration at which 50% of cell growth is inhibited. The inhibitory concentration (IC₅₀) of Standard Cisplatin, Pure Taxifolin and Taxifolin Niosomes were determined using the MTT assay. IC₅₀ of Standard Cisplatin, Pure Taxifolin in MCF-7 cells was found to be 88.08 µg/ml and 75.19 µg/ml correspondingly. An IC₅₀ of 66.65 µg/ml for Taxifolin optimised niosomes formulation (FC6) demonstrated that the cytotoxic effect of Taxifolin loaded formulations was significantly greater than that of free Taxifolin. The results of this work provide credence to theory that niosomes, by virtue of their capacity to encapsulate phytochemical components and improve their efficacy with time, could be a useful nanocarrier for cancer treatment.⁴³

CONCLUSION

It was shown that the cytotoxic effect of Taxifolin loaded formulations was much larger than that of free Taxifolin, with an IC₅₀ of 66.65 µg/ml for the Taxifolin optimised niosomes formulation (FC6). In conclusion, this study's findings support the idea that niosomes are a capable nanocarrier for cancer management due to their capability to excellently encapsulate phytochemical components and enhance their efficacy over time.

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