

Synthesis, Characterization, and Antiproliferative Evaluation of a Quinazoline-Based Compound Against A549 Non-Small-Cell Lung Carcinoma.

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ABSTRACT

A novel quinazoline-based compound, 4-(1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methylbenzoic acid, was synthesized and characterized as a potential therapeutic agent targeting non-small-cell lung carcinoma (NSCLC) cells. The compound was prepared via a two-step synthetic process—alkylation followed by condensation—and its structure was confirmed through infrared (IR), nuclear magnetic resonance (NMR), mass spectrometry, and X-ray diffraction analyses. To evaluate its antiproliferative efficacy, a concentration-dependent cytotoxicity assay (MTT) was conducted using A549 human NSCLC cells. The compound exhibited strong cytotoxic potential with a calculated IC value of 19.88 µg/mL, alongside significant apoptotic induction confirmed by acridine orange/ethidium bromide (AO/EtBr) dual staining. Morphological assessment under microscopy revealed pronounced apoptotic features that increased in a time-dependent manner, with 57% and 66% apoptotic cells present at 24 and 48 hours, respectively. Molecular docking and induced fit docking studies against the KRAS^{G12C} protein (PDB: 7AIW) revealed high-affinity binding, stabilized by a network of hydrogen bonds and hydrophobic interactions involving critical active-site residues, reflected in a substantial MMGBSA binding energy (−39.25 kcal/mol) and an excellent docking score (−4.47). These findings demonstrate that the synthesized quinazoline derivative possesses promising anticancer activity mediated by apoptosis and strong interactions with oncogenic targets, highlighting its potential as a lead molecule for further mechanistic exploration and drug development against NSCLC.

Keywords: Synthesis Quinazoline derivative, A549 cell line, Cytotoxicity; Apoptosis, Insilico Studies

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INTRODUCTION

Lung cancer remains the leading cause of cancer-related mortality globally, contributing significantly to the overall cancer burden. Among its subtypes, non-small-cell lung carcinoma (NSCLC) accounts for approximately 85% of all cases, characterized by aggressive proliferation, metastasis, and resistance to standard therapies[1,2]. Despite recent advances in targeted and immunotherapies, the five-year survival rate for advanced NSCLC remains discouragingly low. The persistent challenges of multidrug resistance and systemic toxicity necessitate the continued search for novel therapeutic scaffolds with selective cytotoxicity and improved pharmacodynamic performance.

Heterocyclic compounds, particularly those based on the quinazoline nucleus, have emerged as potent pharmacophores in anticancer drug discovery. Quinazoline derivatives are well known for their ability to inhibit tyrosine kinases, modulate apoptosis-related pathways, and influence cell cycle arrest[3,4]. Many clinical antineoplastic agents, including gefitinib and erlotinib, owe their efficacy to the inhibitory function of

quinazoline cores on the epidermal growth factor receptor (EGFR)[5]. In this context, hybrid quinazoline-based structures incorporating heteroaromatic rings and substituted alkyl chains have shown promising roles in mediating cytotoxic activity through ROS generation, mitochondrial disruption, and kinase pathway modulation[6,7].

synthesized as a substituted quinazoline derivative, exemplifies this class's structural adaptability and pharmacological potential. Its design integrates a 6,7-dimethoxyquinazoline scaffold with a substituted benzoic acid derivative through a piperidyl linkage, providing diverse electronic and hydrophobic interactions conducive to target affinity[8]. Spectroscopic and crystallographic characterizations confirmed the compound's structure and purity, while computational docking studies predicted its binding propensity with kinase targets associated with NSCLC proliferation. Investigating its cytotoxic potential against A549 lung carcinoma cells provides a vital link between the compound's molecular architecture and biological activity, enabling the assessment of its

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mechanistic influence on cancer cell viability and apoptosis[9,10]

The KRAS^{G12C} mutation, a common oncogenic alteration in non-small-cell lung carcinoma (NSCLC), plays a critical role in driving tumor development by continuously activating cell signaling pathways responsible for cell growth and survival. This mutation involves the substitution of glycine with cysteine at position 12 of the KRAS protein, which locks it in an active state, triggering cascades such as MAPK and PI3K that promote uncontrolled proliferation. Representing about 10-15% of NSCLC cases, particularly lung adenocarcinoma, KRAS^{G12C} mutation is often linked to smoking and distinct co-mutations that influence disease progression and treatment outcomes. The PDB structure 7AIW corresponds to this mutant KRAS in a conformation ideal for molecular docking studies targeting the unique cysteine residue, allowing for selective inhibition efforts. Targeting KRAS^{G12C} has become a promising therapeutic strategy due to its direct involvement in NSCLC pathogenesis and the availability of novel covalent inhibitors that exploit this mutation, aiming to improve clinical responses in patients resistant to conventional therapies. Selecting the KRAS^{G12C} PDB 7AIW structure enables precise evaluation of drug binding and interaction dynamics relevant to this aggressive and therapeutically challenging subtype of NSCLC.

This study aimed to (synthesize and characterize through modern spectroscopic methods, evaluate its cytotoxic potential in the A549 NSCLC cell line, and analyze its mechanism of action

through biochemical and morphological assessments. Integrating the synthetic, spectroscopic, and biological findings presents a comprehensive evaluation of the pharmacological potential of this newly developed quinazoline derivative[11]. Despite advances in targeted therapies for non-small-cell lung carcinoma (NSCLC), challenges such as drug resistance and limited efficacy continue to hinder successful treatment outcomes. Quinazoline derivatives have shown substantial promise as anticancer agents, particularly due to their capacity to inhibit kinase-driven pathways involved in tumor growth and survival. Building on this knowledge, the present study focuses on the synthesis and characterization of a novel quinazoline-based compound designed to target NSCLC. We aim to evaluate its cytotoxicity and apoptotic potential in A549 lung carcinoma cells, alongside computational docking studies to elucidate its molecular interactions with relevant protein targets. This integrated approach seeks to provide insights into the compound's therapeutic feasibility and mechanism of action in NSCLC treatment

Materials and Methodology Chemicals and Reagents

All chemicals and solvents used were of analytical grade and procured from Sigma-Aldrich and Merck. A549 NSCLC cells were obtained from the National Centre for Cell Science (NCCS, Pune, India). Dulbecco's Modified Eagle Medium (DMEM), fetal bovine serum (FBS), MTT reagent, phosphate-buffered saline (PBS), and trypsin-EDTA were purchased from Thermo Fisher Scientific[12].

Experimentalsection

Step1:Alkylationof1,4-dihydropyridinederivative

A solution of 1,4-dihydropyridine (100mg, 0.60mmol) in 0.5M dioxane was treated with 4-(bromomethyl)benzoic acid (76mg, 0.63mmol). The reaction mixture was refluxed for 8 hours under constant stirring. After the completion of the reaction, confirmed by TLC, distilled water was added, and the mixture was further heated for 2 hours[13]. The organic layer was separated using diethyl ether, dried over anhydrous MgSO₄, and the solvent was evaporated under reduced pressure. The crude product was purified by column chromatography using hexane/ethyl acetate (6:1) as eluent to yield the alkylated intermediate.

Step2:Condensationwith4-chloro-6,7-dimethoxyquinazoline

The intermediate was reacted with 4-chloro-6,7-dimethoxyquinazoline (3.14 g, 13.98 mmol) and 3-(4-piperidyl)propanoic acid (2.0 g, 12.72 mmol) in isopropanol (100 mL). The mixture was stirred at 90°C for 3 hours[14]. The reaction mass was evaporated to dryness and triturated with dichloromethane (20 mL).

Characterization

The compound was characterized using IR spectroscopy (PerkinElmer Spectrum IR, 4000–450 cm⁻¹), ¹H-NMR and ¹³C-NMR (400 MHz, DMSO-d₆ solvent), mass spectrometry (ESI-MS, positive mode), [15] and powder X-ray diffraction (Cu K α source, λ = 1.5418 Å) Cytotoxicity Assay (MTT)

A549 cells were cultured in DMEM supplemented with 10% FBS, 1% penicillin–streptomycin, and incubated at 37°C under 5% CO₂. Cells were seeded at a density of 1 × 10⁴ cells per well in a 96-well plate and incubated overnight. 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid was dissolved in DMSO and diluted to final concentrations ranging from 5 to 100 μ M. After 24h treatment, 10 μ L of MTT reagent (5mg/mL) was added and incubated for 4 hours[16,17]. The formazan crystals were dissolved in DMSO (200 μ L/well), and absorbance was measured at 570 nm using a microplate reader. The cell viability percentage was calculated relative to untreated control, and IC₅₀ was derived via nonlinear regression.

Morphological Examination

A549 cells treated with 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid (10 and 25 μ M) were inspected under an inverted phase-contrast microscope to observe cellular shrinkage, rounding, and detachment, indicative of apoptotic morphology.

Apoptotic staining AO/ETBR

The acridine orange/ethidium bromide (AO/EtBr) dual staining technique was performed to identify apoptotic

changes in A549 cells after treatment with the test compounds. Approximately 3×10^4 cells were seeded in six-well plates and incubated at 37°C in a humidified CO_2 atmosphere until they reached 70–80% confluence. The cells were then treated with the IC_{50} concentration of each compound for 24 and 48 hours, while untreated cells served as the control. After the incubation period, the medium was carefully removed, and the wells were washed twice with Dulbecco's phosphate-buffered saline (DPBS) to eliminate non-adherent cells. A staining solution containing acridine orange and ethidium bromide at equal concentrations (100 $\mu\text{g}/\text{mL}$ each) was freshly prepared, and 20 μL of this mixture was added to each well [18]. The plates were incubated for 30 minutes at room temperature in the dark to avoid photo bleaching, followed by gentle PBS washing to remove excess dye. The cells were immediately examined under a fluorescence microscope fitted with appropriate filters, and images were captured for analysis. Green fluorescent cells were considered viable, yellow to orange cells indicated early apoptosis, and red cells represented late apoptotic or necrotic stages. The proportion of apoptotic cells was quantified by counting stained cells in random microscopic fields and calculating the percentage of apoptotic cells relative to the total number of cells. All procedures were carried out in triplicate to ensure accuracy and reproducibility.

In silico Studies

The molecular docking studies were conducted using the KRASG12C protein structure available from the Protein Data Bank (PDBID: 7AIW). The protein was prepared in Maestro using the Protein Preparation Wizard: this involved assignment of bond orders, addition of hydrogens, and optimization of hydrogen-bonding networks. Missing side-chains and loops were generated as required, and water molecules beyond 5 Å from the binding site were removed to ensure optimal docking conditions. The structure was then minimized using the OPLS4 force field to relieve any local strain and optimize geometry. The ligand was designed using a 2D sketch in Maestro and subjected to Lig Prep for generating appropriate ionization states (at $\text{pH} 7.0 \pm 2.0$), tautomers, and energy minimization via OPLS4. Prior to docking, the receptor grid was generated around the KRASG12C active site, focusing on the region encompassing residue Cys12, Gly13, and the neighboring catalytic pocket. Docking simulations were performed using the Glide XP (Extra Precision) algorithm. Multiple ligand conformations were sampled, and the best pose based on the lowest docking score was selected for further analysis. The binding free energy of the protein–ligand complex was calculated using the Prime MMGBSA method, with ΔG_{bind} estimated in vacuum to evaluate the stability and affinity of the interaction. All fixed and flexible receptor/ligand options were properly configured throughout the process to ensure accurate predictions of binding energies and pose selection. Induced fit docking

Induced fit docking (IFD) was implemented using the Schrodinger Maestro suite, enabling both receptor and

ligand flexibility for enhanced accuracy of binding predictions. The process began with protein structure preparation (PDBID: 7AIW), where hydrogens were added, missing side-chains and loops were rebuilt, and protonation states were adjusted. Hydrogen bonds were optimized, and the resulting protein structure was minimized to remove local strain. The ligand was constructed in 2D using Maestro's sketch tool and processed by LigPrep for generation of correct tautomers, ionization states, and conformational sampling at physiological pH. The active site for docking was defined using grid generation centered on the target region of KRASG12C [19]. The IFD protocol was run as follows: first, Glide standard precision (SP) docking was carried out to generate initial ligand poses with softened van der Waals potentials. For the top candidate poses, near by protein side-chains were sampled and refined, allowing them to adjust in response to ligand binding. The protein–ligand complexes were then minimized using the OPLS4 force field. After refinement, the ligand was re-docked into the newly optimized active site using Glide XP (extra precision) docking. The best binding poses were selected according to the XP docking scores. Finally, binding free energies (ΔG_{bind}) were calculated using the Prime MMGBSA method under vacuum conditions, thus affording a rigorous assessment of the affinity and stability of the ligand–protein interaction after full consideration of induced fit effects [20–22].

RESULT AND DISCUSSION

The synthesis of 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid was accomplished via a two-step sequence involving alkylation and subsequent condensation reactions. Initially, 1,4-dihydropyridine (100 mg, 0.60 mmol) was subjected to nucleophilic substitution with 4-(bromomethyl)benzoic acid (76 mg, 0.63 mmol) in dioxane (0.5M) under reflux conditions for 8 hours. The subsequent hydrolysis step involved the addition of water and heating for an additional 2 hours to complete the reaction. Isolation of the alkylated intermediate was achieved through extraction with diethyl ether, washing, drying over MgSO_4 , and concentration under reduced pressure. Purification by column chromatography (hexane/ethyl acetate, 6:1) afforded the intermediate in acceptable yield.

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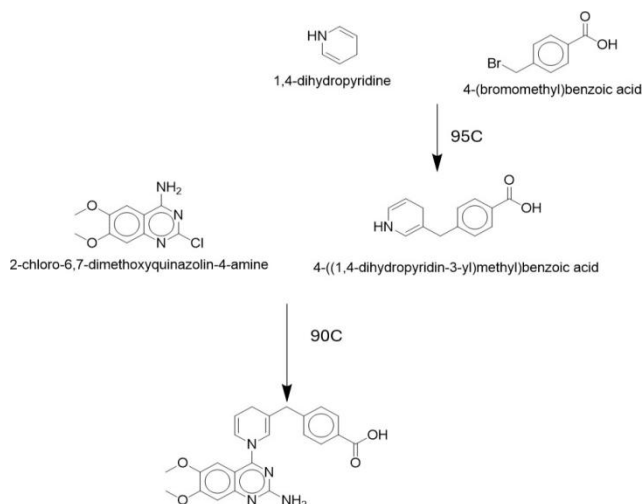


Figure 1: Scheme 14—((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid

In these second step, the purified intermediate underwent nucleophilic aromatic substitution with 4-chloro-6,7-dimethoxyquinazoline (3.14 g, 13.98 mmol) and 3-(4-piperidyl)propanoic acid (2.0 g, 12.72 mmol) in isopropanol at 90 °C for 3 hours. Workup involved evaporation and trituration with dichloromethane, yielding 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid as a white solid in moderate yield (42%)

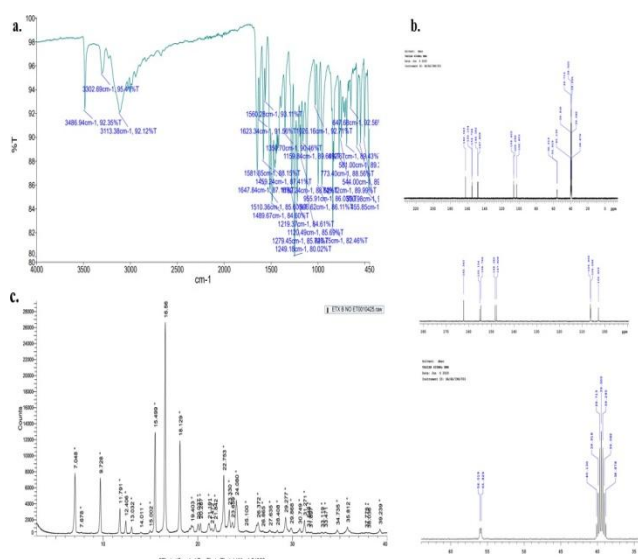


Figure 2 Spectral confirmation of 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid by IR, NMR and XRD data

the structure of 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid was elucidated via a combination of infrared, nuclear magnetic resonance, and mass spectral analysis as presented. The IR spectrum revealed key absorptions at 3486 cm^{-1} for O–H stretching, 3302 cm^{-1} for N–H stretching, and 1648 cm^{-1} assignable to carbonyl and imine groups, thus supporting the quinazoline framework. The ^1H NMR spectrum showed aromatic protons spanning δ

7.93–7.06 ppm, methoxy group signals at δ 3.86–3.88 ppm, and aliphatic proton resonances characteristic of the dihydropyridine segment between δ 2.50–3.32 ppm. The ^{13}C NMR spectrum further confirmed the structure with resonances for aromatic carbons in the δ 162.3–147.8 ppm region and methoxy-substituted carbons near δ 56.0 ppm. Finally, mass spectrometry displayed a molecular ion peak at m/z 540.39, matching the expected molecular mass for this compound, thus conclusively affirming its identity. X-ray diffraction (XRD) analysis further confirmed the structural identity and crystallinity of 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid. The powder XRD spectrum displayed numerous sharp and intense peaks, indicative of a well-defined crystalline phase. The characteristic 2θ values observed included 7.048°, 9.728°, 11.791°, 15.499° (major peak), 16.569°, 18.129°, 22.753°, 27.472°, and 35.812°, among many others. The occurrence and distribution of these peaks, with the strongest reflection at 16.569°, affirm the presence of a unique crystalline structure corresponding to the synthesized compound it was depicted on Figure 2c.

The two-step synthetic approach provided a viable route to access the quinazoline-based scaffold with reasonable overall yield and satisfactory purity as evidenced by spectral data. The regioselective alkylation and subsequent displacement on the chloroquinazoline in the presence of the piperidyl acid were achieved under relatively mild conditions, demonstrating synthetic feasibility for the preparation of analogues in future structure-activity relationship studies. The cytotoxic potential of 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid was evaluated on the human lung carcinoma A549 cell line using the MTT colorimetric assay. The compound exhibited a clear, concentration-dependent inhibitory effect on cell viability. As the concentration increased from 1 $\mu\text{g/mL}$ to 512 $\mu\text{g/mL}$, a significant reduction in viable cell percentage was observed—from 91.46% at the lowest concentration to 5.49% at the highest. The calculated IC_{50} value was 19.879 $\mu\text{g/mL}$, indicating substantial cytotoxic potency (Figure 3).

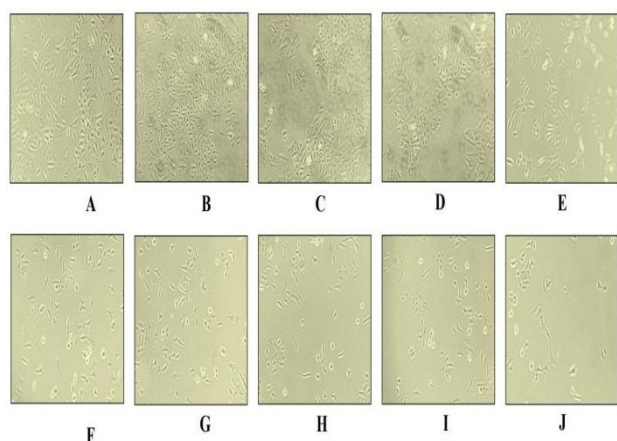


Figure 3. Dose–response curve showing the cytotoxic effect of the synthesized quinazoline derivative on A549 lung carcinoma cells (MTT assay).

The steep decline in cell viability with increasing concentrations suggests that 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-

yl)methyl)benzoic acid effectively interferes with mitochondrial metabolic activity, leading to cell death. Compared to the control group, treated cells showed pronounced morphological alterations such as rounding, detachment, and cell shrinkage, all of which are characteristic of apoptotic processes. The regression analysis ($y = -13.84 \ln(x) + 91.377$) confirmed a strong logarithmic relationship between concentration and cytotoxic response (Table 1). Further evaluation using acridine orange/ethidium bromide (AO/EtBr) dual staining revealed a progressive increase in apoptotic cell population with time. After 24 hours of treatment, approximately 57% of the cells exhibited apoptotic morphology, which further increased to 66% after 48 hours. Illustrates on Figure 4 that 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid triggers apoptosis rather than necrotic cell death.

Table 1-logarithmic relationship between concentration and cytotoxic response

Concentration ($\mu\text{g/mL}$)	% Viability	Duplicate (%)	Triplicate (%)	Mean Viability (%)	Standard Deviation
512	5.628	5.472	5.380	5.493	± 0.125
256	16.021	15.540	15.632	15.731	± 0.255
128	24.667	24.694	24.997	24.786	± 0.183
64	32.502	32.870	32.518	32.630	± 0.207
32	41.645	41.846	41.937	41.809	± 0.149
16	52.260	52.038	52.022	52.106	± 0.132
8	61.895	62.485	62.306	62.229	± 0.302
4	72.023	72.888	73.267	72.726	± 0.637
2	83.064	82.832	83.064	82.987	± 0.134
1	91.137	91.440	91.802	91.460	± 0.332

Collectively, the MTT and AO/EtBr analyses indicate that 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid possesses potent anticancer activity against A549 cells by inducing apoptosis in a dose- and time-dependent manner. The relatively low IC_{50} value compared to other tested compounds supports its potential as a promising candidate for further mechanistic and therapeutic investigations.

In silico Studies

The molecular docking and induced fit docking (IFD) studies revealed detailed insights into the specific residue interactions responsible for anchoring the ligand within the KRAS G12C binding pocket. The ligand formed stabilizing hydrogen bonds and polar contacts primarily with Asp12, Gly13, and Gly60, which are known to be critical for recognition within the active site. Additional hydrophobic and π - π stacking interactions were observed with Val9, Thr58, Gln61, His95, Tyr96, and Ala85.

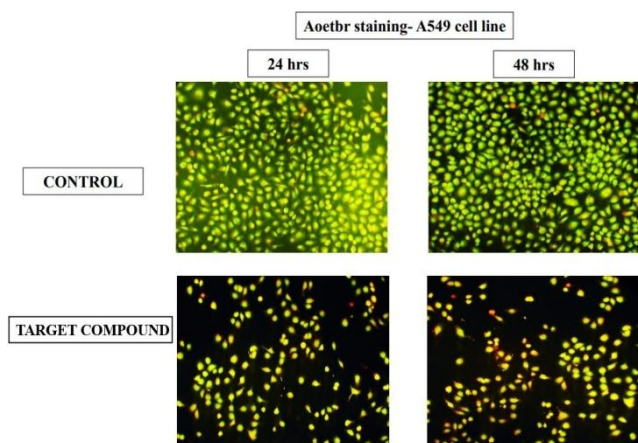


Figure 4. Fluorescence microscopic images of AO/EtBr-stained A549 cells after 24 and 48 hours of compound treatment showing viable (green), early apoptotic (yellow/orange), and late apoptotic (red) c

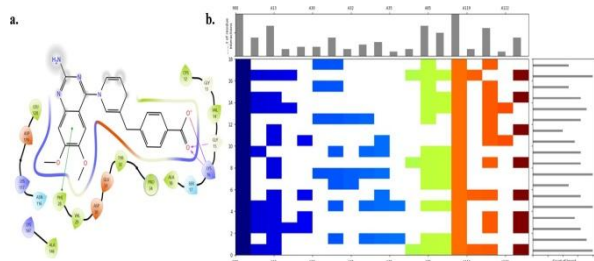


Figure 5 a. 2D interaction diagram of 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid and **b.** heatmap showing key residues involved in ligand binding, highlighting frequent hydrogen bonding and hydrophobic interactions.

Notably, the interaction profile also involved residues such as Ala13, Ala30, Ala32, Ala35, Ala119, and Ala122, as confirmed by both the interaction diagram and contact heatmap. These interactions represent a combination of hydrogen bonds, van der Waals contacts, and aromatic stacking, collectively contributing to the strong binding energy (MMGBSA-39.25 kcal/mol) and docking score (-4.468) observed in computational studies. The spatial orientation of the ligand favors optimal engagement with both polar and apolar regions of the KRASG12C active site, supporting its candidacy for further optimization as a targeted inhibitor

CONCLUSION

The newly synthesized quinazoline derivative, 4-((1-(2-amino-6,7-dimethoxyquinazolin-4-yl)-1,4-dihydropyridin-3-yl)methyl)benzoic acid, demonstrated potent cytotoxic and apoptotic effects against A549 human lung cancer cells. Structural confirmation through spectroscopic analyses validated the successful formation of the target compound. The MTT assay results established a clear dose-dependent reduction in cell viability, highlighting its significant antiproliferative potential. Furthermore, AO/EtBr dual staining analysis revealed that the compound promotes apoptosis in a time-dependent manner, suggesting that its anticancer activity is primarily mediated through

apoptotic pathways rather than necrosis. The combination of synthetic feasibility, strong cytotoxic response and confirmed apoptotic induction under scores the pharmacological promise of this quinazoline scaffold. Future studies focusing on molecular docking, target interaction analysis, and in vivo evaluation are warranted to elucidate its mechanism of action and therapeutic applicability in NSCLC treatment

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