

Targeted Bisphosphonate Application as an Adjunctive Treatment to Scaling and Root Planing

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Abstract

Bisphosphonates are potent antiresorptive agents widely used in the management of osteoporosis, Paget's disease of bone, osteogenesis imperfecta, and skeletal complications associated with malignancies. Their strong affinity for hydroxyapatite and ability to suppress osteoclast-mediated bone resorption make them clinically valuable; however, these same properties have important implications for dental practice. This review examines the pharmacological characteristics, classification, mechanisms of action, pharmacokinetics, pharmacodynamics, and major clinical applications of bisphosphonates, with particular emphasis on their role in dentistry. Bisphosphonates inhibit osteoclast activity through distinct mechanisms, with nitrogen-containing agents primarily targeting farnesyl pyrophosphate synthase within the mevalonate pathway, thereby reducing osteoclast function and bone turnover. In periodontal therapy, locally or systemically administered bisphosphonates have been investigated as adjuncts to scaling and root planing, with evidence indicating potential improvements in periodontal parameters, including reductions in probing pocket depth. Their localized application may provide therapeutic benefits while minimizing systemic adverse effects. Bisphosphonates have also demonstrated potential benefits in implant therapy through localized surface coatings that may enhance osseointegration and preserve marginal bone. Conversely, systemic bisphosphonate therapy may influence orthodontic tooth movement by suppressing bone remodeling. A major dental concern is medication-related osteonecrosis of the jaw (MRONJ), particularly following invasive dental procedures in patients receiving potent intravenous bisphosphonates. Therefore, although bisphosphonates offer significant therapeutic benefits for skeletal and periodontal conditions, their use requires careful dental assessment, individualized risk evaluation, and appropriate preventive management. Understanding their dual therapeutic and adverse effects is essential for safe and effective multidisciplinary patient care.

Keywords: Bisphosphonates; Scaling and Root Planing; Periodontitis; Osteoclast Inhibition; Bone Remodeling; Periodontal Therapy

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INTRODUCTION

Bisphosphonates (BPs) around a century ago were first synthesized as antiscaling compounds by German chemists.¹ Since then, the BPs have evolved to become one of the most successful and widely used groups of drugs in the world, used for improving the treatment of Paget's disease of bone, osteoporosis, metastatic bone disease, multiple myeloma^{2,3} Apart from their use for treatment of bony diseases, they are used in non-skeletal applications such as neurodegenerative diseases as well.⁴ Recently they are used as bone-targeting carriers for other drugs such as antibiotics, hormones and anti-cancer drugs⁵⁻⁷

Structurally, bisphosphonates consist of two phosphonate groups linked by a carbon atom, the general chemical structure of bisphosphonates, with the attached carbon chains (denoted as R and R'), varying depending on the specific type of bisphosphonate is shown in figure 1.⁸

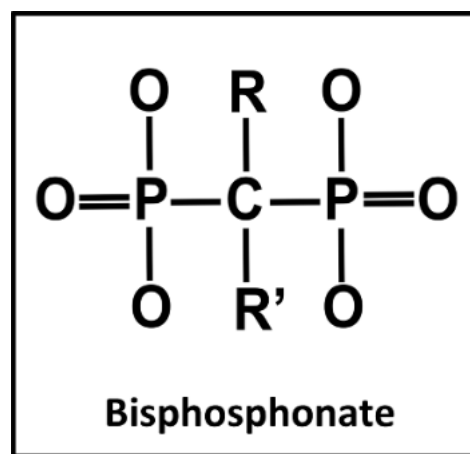
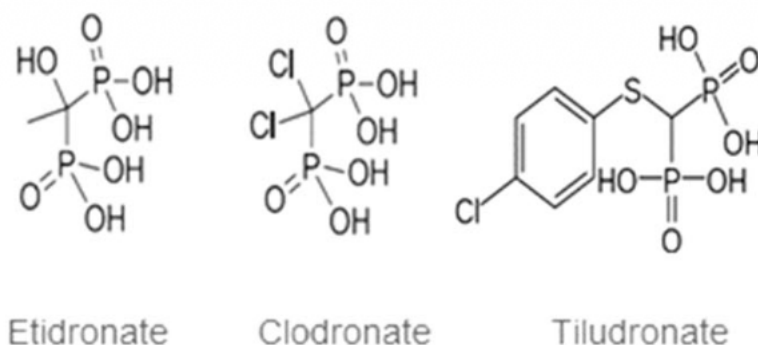


Figure 1: Chemical structure of bisphosphonates. The “R” and “R'” refers to long carbon chains that differ depending on the type of bisphosphonate.

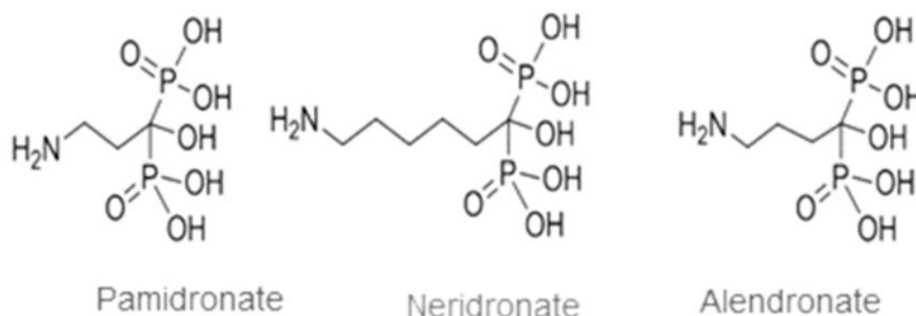
CLASSIFICATION

These drugs are divided into first-generation non-nitrogen-containing (clodronate, etidronate and tiludronate), second and third generation nitrogen-containing (alendronate, risedronate, ibandronate and zoledronate) where third generation differ from the others because they adhere more tightly to hydroxyapatite mineral in bone⁹

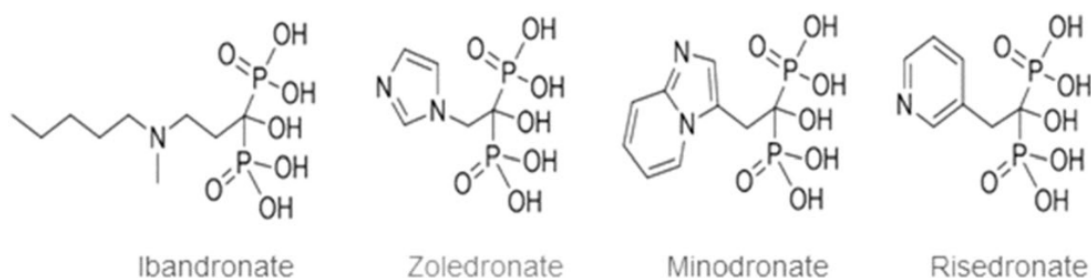
1st generation: Non nitrogen containing



2nd generation: nitrogen containing with short alkyl chains



3rd generation: nitrogen containing with branch or ring structure



MECHANISM OF ACTION

All bisphosphonates inhibit bone resorption particularly in areas with active resorption by attaching to hydroxyapatite binding sites on the bone. As osteoclasts resorb bone, the bisphosphonate embedded in the bone is released and impairs the osteoclast's ability to continue bone resorption.⁵ One of the novel actions of bisphosphonates has been related to preservation of osteoblasts and osteocyte viability.¹⁰

Nitrogen-containing

bisphosphonates (2nd and 3rd generation) work by inhibiting farnesyl pyrophosphate synthase, which is important in promoting attachment of the osteoclast to the bone. As a result, the osteoclast detaches from the bone surface, thus inhibiting bone resorption.¹¹

Non-nitrogen-containing

bisphosphonates, on the other hand, are metabolized within the cell to substrates that replace the terminal pyrophosphate moiety of adenosine triphosphate, leading to forming of a nonfunctional molecule that competes with adenosine triphosphate in the energy metabolism of the cell. Due to this osteoclast apoptosis is inhibited, in

turn leading to an overall decrease in bone breakdown.¹¹

So, overall their mechanism of action can be divided into two primary pathways: inhibition of osteoclast activity and reduction of bone turnover.

1. Inhibition of Osteoclast Activity

Bisphosphonates exert their effects by disrupting the function of osteoclasts, the cells responsible for bone resorption. The nitrogen-containing bisphosphonates with its effect on inhibition of farnesyl pyrophosphate synthase (FPPS), a key enzyme in the mevalonate pathway leads to the impairment of the osteoclast's ability to adhere to the bone surface. As the osteoclasts do not adhere to bone, no ruffled border is formed which is an essential step for bone resorption.¹³ Additionally, bisphosphonates induce programmed cell death in osteoclasts, reducing their population and activity.¹⁴

2. Reduction in Bone Turnover

Bisphosphonates inhibit osteoclast-mediated bone resorption, which can lead to a shift in the bone remodeling process. Moreover, bisphosphonates bind to hydroxyapatite, preventing mineral loss and reducing bone brittleness.¹⁵

PHARMACOKINETICS

All oral BPs have a low and highly variable bioavailability, where approximately 99% of any orally administered BP are excreted unchanged into the faeces. This increases to 100% if the drug is ingested with calcium- or magnesium-containing foods, drinks or drugs.¹⁶ Oral BPs are taken up in the stomach, duodenum and ileum by means of paracellular and active transport.^{17,18} However, when co-ingested with gastric pH-raising drugs such as ranitidine and omeprazole its bioavailability increases.¹⁶ Also enhancers such as caproic acid increase absorption of BPs.¹⁹

In the circulation, BPs remain bound to plasma and serum proteins with lower binding in plasma which has been attributed to endogenous displacers. Moreover, this binding is dependent on the type of BP, its concentration, pH, calcium.¹⁷

BPs without any first phase or second phase metabolism go in the liver. The only pathway of metabolism for BPs is intracellular transformation to cytotoxic ATP analogues of the non-nitrogen-containing BPs etidronate, clodronate and tiludronate.²⁰ Their elimination is via the kidneys or they are deposited in the calcified tissue. And they are eliminated predominantly through glomerular filtration, with the renal clearance of the drug correlating closely with creatinine clearance.²¹⁻²³ None of the FDA-approved BPs have been approved for use in patients with a creatinine clearance below 35 mL/min. The general recommendation for the use of BPs in patients on renal replacement therapy is to decrease the BP dose by 50%, increase the infusion time of intravenously administered BPs, and limit the overall number of doses provided.²⁴

BPs bind to hydroxyapatite crystals in bone²⁵ more strongly at sites of new mineral deposition at the "calcification front" in osteoid. They also bind well to resorption sites, presumably because calcium phosphate minerals are exposed during resorption. BPs less avidly bind to sites

with little bone turnover.^{26,27} In the skeleton, more BP is taken up by trabecular bone than cortical bone, it's because of the higher rate of turnover and greater surface area available in trabecular bone.²⁸ Thus there is a non-homogeneous distribution of BPs within the skeleton.²⁸

The distribution of BP to bone seems to differ from BP to BP.^{26,27} Some BPs bind more strongly to hydroxyapatite than others. This difference in BP binding properties were shown to determine not only the extent to which a single BP dose would bind to the skeleton, but also how well BPs penetrate into the canalicular network of bone. In general, however, BPs with weaker bone-binding properties such as risedronate appear to penetrate deeper into the bone than stronger binding BPs such as alendronate.²⁹

After binding to the bone, BPs are either released again by dissolution, or are covered within new bone that overlays resorption sites for later release into the circulation during osteoclast-mediated bone resorption, either directly or via osteoclasts.^{26,27}

Intracellularly, the nitrogen (N)-containing BPs inhibit farnesyl pyrophosphate synthase (FPPS), which ultimately leads to the loss of function of the osteoclasts.³⁰ Some of the BP is released into the circulation during resorption from bone, and some may be released after apoptosis and death of osteoclasts. BPs released from bone may undergo re-uptake onto bone surfaces. Thus, these BPs are detected in urine for years after treatment is discontinued.³¹

PHARMACODYNAMICS

A landmark discovery in the 1990s, was made about the first clinical use of bisphosphonates, and how BPs act biochemically within cells. For the N-containing BPs the inhibition of osteoclast-mediated bone resorption is mediated via inhibition of the enzyme, farnesyl pyrophosphate synthase (FPPS), an essential enzyme of the mevalonate pathway and inhibition of this enzyme leads to a decrease in farnesyl pyrophosphate, and geranyl geraniol pyrophosphate (GGPP), leading to a decreased

prenylation of signal transduction proteins such as Rac, Ras and Rho, which in turn leads to a loss of function of the osteoclasts.^{30,31-34} In addition to osteoclasts, BPs have also been shown to work on other bone cells, osteoblasts as well as osteocytes, by preserving their viability.¹⁰ Some BPs also seem to have direct anti-tumour activity,^{35,36} eg Zoledronate might exert its anti-tumour effects through the miR21/PTEN/Akt signalling and Activin signalling pathways.^{37,38}

The potential actions of BPs in other indications such as for neurodegenerative diseases are not yet fully explored. Their ability to inhibit calcium crystal growth as well as to inhibit FPP synthase in the mevalonate pathways has been postulated.³⁹

The most important clinical outcome parameters for BP treatment are reduction of fracture risk in osteoporosis, skeletal-related events (SREs) in metastatic bone disease and more disease-specific parameters in other diseases such as pain and deformation in Paget's disease of bone, or number of fractures and bone pain in osteogenesis imperfecta.⁴⁰

ROLE OF BISPHOSPHONATES IN CLINICAL PRACTICE

1. Osteoporosis

Osteoporosis is characterized by a decrease in bone mass and a deterioration in the architecture of the bones, which leads to an enhanced fragility of the skeleton and a greater risk of fracture. It is defined as present in women when the bone mass is more than 2.5 SD below that of the young woman (t score).⁴¹

The clinical manifestations of osteoporosis are fractures, which occurs often spontaneously or after minimal trauma, and their consequences. Osteoporosis is diagnosed and assessed quantitatively by techniques that measure bone mineral density (BMD), most commonly dual X-ray absorptiometry.

Treatment of osteoporosis with bisphosphonate is done. Bisphosphonates prevent the decrease in BMD. With etidronate being the first compound thoroughly investigated,⁴² followed by alendronate⁴³ and later risedronate.⁴⁴ Bisphosphonates are active in osteoporotic women, also they are used for elderly women without osteoporosis⁴⁵. Risedronate therapy prevents and partially even reverses the bone loss in glucocorticoid-treated patients.⁴⁶ All the bisphosphonates induce a marked decrease in bone turnover when given in doses effective on BMD. Both bone formation and resorption are decreased. Both alendronate and risedronate decrease by about

one-half the occurrence of vertebral and nonvertebral fractures in osteoporotic patients.^{47,48} Lastly, bisphosphonates do reduce fractures also in children with osteogenesis imperfecta⁴⁹

The treatment regimens for the three main commercialized compounds are the following:⁴¹

- Alendronate: The dose recommended by the producers is 10 mg orally daily, 5 mg in Japan. It has a similar effect on BMD when given once weekly at 70 mg, so the weekly regimen is used today in many countries.
- Etidronate: The regimen recommended by the producer is 400 mg daily orally for 2 weeks every 3 months.
- Risedronate: The recommended regimen is 5 mg daily orally or 35 mg once weekly.

2. Glucocorticoid-Induced and Transplant-Associated Osteoporosis

Whereas bisphosphonates have become the primary therapeutic choice for treatment of postmenopausal osteoporosis, few recognize that glucocorticoid therapy leads to bone loss. Numerous clinical trials have now determined that bisphosphonates are highly effective at limiting bone losses in patients receiving glucocorticoids or transplants.^{46,50,51}

Multiple studies have documented that both oral and IV bisphosphonate therapy are capable of limiting the bone loss that frequently occurs with either solid organ⁵²⁻⁵⁶ or bone marrow transplant.⁵⁷⁻⁶⁰

3. Bisphosphonate Therapy for Children

Bisphosphonates have become the mainstay of therapy for Osteogenesis Imperfecta (OI). A regimen developed by Glorieux of cyclic IV pamidronate (given in 3-day cycles every 2 to 4 months at an annual dose of 9 mg/kg) has been used most successfully, leading to an 88% increase in cortical thickness, a 46% increase in trabecular bone volume and substantial improvement in functional status.^{61,62}

More recently, several studies have demonstrated that oral alendronate can also lead to substantial increases in BMD in OI affecting children.⁶³⁻⁶⁵

4. Paget Disease of Bone

Bisphosphonates are the cornerstone of therapy for Paget disease of bone, it profoundly suppresses the increased bone resorption underlying the disease. Oral and iv bisphosphonates are all FDA-approved for the treatment of Paget disease of bone replacing previous modalities because of their superior ability to suppress osteoclast activity.⁶⁶⁻⁶⁹

5. Bisphosphonates in Malignancy

- a. Breast Cancer—For patients with breast cancer metastatic to bone, treatment with IV preparations of pamidronate, zoledronic acid and ibandronate has been shown to substantially relieve skeletal pain and reduce skeletal complications. Of the oral nitrogen containing bisphosphonates, only ibandronate (given in a daily dosage of 50 mg) has been effective in reducing bone pain and limiting skeletal complications of breast cancer.⁷⁰⁻⁷⁷
- b. Prostate Cancer – recent studies have shown a role of increased bone resorption in metastatic prostate cancer . Among the bisphosphonates, only zoledronic acid has been demonstrated to reduce skeletal bone–related events in men with hormone-refractory prostate cancer.⁷⁸⁻⁸⁰
- c. Multiple Myeloma- many studies have shown that both pamidronate and zoledronic acid have an important palliative role in reducing the incidence of skeletal bone–related events associated with myeloma and hypercalcemia , these studies provide a substantial base for considering IV bisphosphonates as the center of current therapies to prevent and treat myeloma-associated bone disease.⁸¹⁻⁸³
- d. Renal cell carcinoma-Use of bisphosphonates in renal cell carcinoma, has been demonstrated to delay the onset and progression of skeletal disease which is due to involvement of secondaries in the bone.⁸⁴

BISPHOSPHONATES AND DENTISTRY

1. Dental implants:

A systemic review has been done and it concludes that patients taking BPs presented a statistically significant higher risk of dental implant failure than patients not taking it.⁸⁵ They postulate that it is due to that it may be due to some potential negative effects of BPs such as its ability to inhibits bone remodelling and turnover impairing the osseointegration of dental implants and the repair of microdamage in these patients.⁸⁶ Also sites with a poor bone quality and lack of bone volume may negatively affect the implant failure rates as seen in a previous study.⁸⁷

Also in the systemic review it is emphasised that higher implant failure rates are expected in cases with iv administration⁸⁵ due to a greater potential for side effects with iv doses of bisphosphonates due to their higher dose and potency.⁸⁸

This effect can further be demonstrated due to the fact that BP loads bone and accumulates in bone 142.8 times faster when administered intravenously in comparison to when given orally.⁸⁹

Also the period that patients have been under BP therapy may also play an important role in the survival of implants.⁸⁵ The higher risk of implant failure may also be influenced by the patients' compliance to the BP therapy.⁸⁵

Lastly, the different degrees of potency among different BPs have different potentials to affect bone metabolism and cause adverse effects and that could directly or indirectly have an effect on dental implant survival.⁹⁰

However some of them also have a beneficial effect on implant systems. Bisphosphonates, can be administered systematically or topically, the latter involving localized application, such as coating the surface of dental implants.⁹¹ Studies have shown positive outcomes when bisphosphonates are coated on implants, enhancing osseointegration and

delaying osteoclast apoptosis, ultimately extending the preservation time of marginal bone around implants.⁹²⁻⁹⁴

2. Periodontal treatment:

Periodontitis is the inflammation of the gingiva caused by bacterial flora, which could lead to the destructive of gingival tissues, or both. It can even cause tooth loss, in more severe cases.⁹⁵ Periodontitis is treated via a procedure known as scaling and root planning (SRP), which physically removes the pathogenic microorganisms causing the inflammation.⁹⁵ However, a different approach to periodontal treatment is seen by adjunctive bisphosphonate therapy in the treatment of periodontitis, it is due to the bisphosphonates' ability to inhibit proinflammatory cytokines and the resorption of alveolar bone.⁹⁶⁻⁹⁹ Alternative routes have been used as well such as administration via intravenously or applied locally, in gel-form (often using alendronate specifically) for the purpose of treating periodontitis.¹⁰⁰⁻¹⁰¹ According to several review articles, a vast number of studies with locally applied showed significant improvements to patients' periodontal conditions, resulting in decreased probing pocket depth.^{95,96} Similar results were found for bisphosphonates administered systemically, whether orally or intravenously, with oral bisphosphonates showing greater improvement compared to IVs.⁹⁵ Although both systemic and local bisphosphonates demonstrate effectiveness, locally applied bisphosphonates are preferred due to having fewer side effects.⁹⁶

3. Orthodontics:

Considering that orthodontic forces promote physical and biological events critical to bone renewal, including recruitment of inflammatory cells, formation of new vessels and tissue remodelling, it is possible that the systemic use of bisphosphonates inhibits the osteo-reabsorption process, and as such bisphosphonates concomitantly with

orthodontic treatment influence these events, decreasing tooth movement and delaying orthodontic treatment.^{102,103} Most of the studies agreed that the systemic use of bisphosphonates contributed to decrease orthodontic movement.¹⁰⁴⁻¹⁰⁹ The reason for the inhibitory effect of bisphosphonates on orthodontic movement can be attributed to disrupting osteoclast function and maintaining the compression sites of the periodontal ligament, where bone resorption becomes necessary for tooth movement to occur.¹⁰⁸

However, in a study by Seifi et al , he reported that despite not observing the influence of using bisphosphonate on orthodontic movement, still found that experimental bisphosphonate significantly inhibited the bone and root resorption in rats, assigning a protective and also beneficial tissue effect to this drug.¹¹⁰

Variations in the drug dose also exerted an influence on the results of these studies , with larger doses associated with smaller and slower orthodontic movements, and smaller doses associated with larger and faster movements.¹⁰⁶

CLINICAL CONCERNS ASSOCIATED WITH BISPHOSPHONATE THERAPY

1. Osteonecrosis of the Jaw

Medication-related osteonecrosis of the jaw (MRONJ) is a serious side-effect in the therapeutic use of AR drugs such as bisphosphonates .¹¹¹ The majority of MRONJ cases occurred in patients receiving intravenous administrations of nitrogen-containing bisphosphonates due to metastatic bone lesions.

Dental infection, teeth extraction and poor oral hygiene are main risk factors for the development of MRONJ.^{112,113} Several investigations propose that common infectious diseases e.g. preexisting pathological inflammatory conditions like baseline osteomyelitis and periapical periodontitis are also contributing factors to

the development of bisphosphonate related osteonecrosis of jaw (BRONJ) following tooth extraction.¹¹⁴⁻¹¹⁷

2. Atrial Fibrillation

In addition to the concern for BRONJ, another concern with bisphosphonate therapy, which has recently come to light, is atrial fibrillation. In the HORIZON Pivotal Fracture Trial, in patients were treated annually with IV zoledronic acid, and a statistically significant increase in the incidence of serious atrial fibrillation was noted.¹¹⁸

3. Over-suppression of Bone Turnover

Because bisphosphonates inhibit osteoclast activity, there has been some concern that prolonged bisphosphonate treatment leads to “frozen bone,” characterized by over suppression of bone remodelling, an impaired ability to repair skeletal microfractures, and increased skeletal fragility.¹¹⁹

4. Hypocalcemia

Hypocalcemia after mostly after iv bisphosphonate administration and can occur in patients with high rates of osteoclast-mediated bone resorption¹²⁰⁻¹²¹

5. Acute Inflammatory Response

Approximately 10% to 30% of patients receiving their first nitrogen-containing bisphosphonate infusion will experience an acute phase reaction, most commonly characterized by transient pyrexia with associated myalgias, arthralgias, headaches, and influenza-like symptoms. This rate declines by more than half with each subsequent infusion, such that a rate of 2.8% was found after the third infusion in the HORIZON trial.¹¹⁸

6. Severe Musculoskeletal Pain

Although all oral and IV bisphosphonate preparations list musculoskeletal pain as a potential adverse effect in their prescribing information, the US FDA recently issued an alert highlighting the possibility of

severe, incapacitating musculoskeletal pain that can occur at any point after initiation of bisphosphonate therapy.¹²²

CONCLUSION

In clinical dentistry, bisphosphonates represent a double-edged sword simultaneously being a systemic bliss and a localized bane. They function as a profound medical bliss by preserving skeletal integrity, preventing debilitating fractures, and managing bone malignancies through the powerful inhibition of osteoclastic bone resorption. However, they transform into a formidable dental bane because this same mechanism halts the high-turnover bone remodelling necessary for oral healing, occasionally triggering Medication-Related Osteonecrosis of the Jaw (MRONJ). Thus overall we can conclude that bisphosphonates are a highly effective medical necessity that requires strict dental caution.

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