

RESEARCH ARTICLE

Box-Behnken Design for Nephrolithiasis: Developing Folic Acid-Chitosan Nanoparticles from *Musa acuminata* Stem Water

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ABSTRACT

Nephrolithiasis, or kidney stones, is a widespread condition causing significant discomfort and health issues. This analysis delves into the development of folic acid-chitosan nanoparticles (FA-CNPs) using *Musa acuminata* stem water to address nephrolithiasis. A Box-Behnken design was used to refine synthesis variables—chitosan concentration, folic acid loading, and reaction time. The study systematically analyzed their effects on nanoparticle size, encapsulation efficiency, and release profile. The optimized FA-CNPs demonstrated potential for improved targeted treatment of kidney stones, highlighting advancements in nanomedicine for effective nephrolithiasis management.

Keywords: Biocompatibility, Antioxidant, Kidney, nutrients, Nanoparticles.

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INTRODUCTION

Nanoparticles enhance drug delivery and bioavailability due to their size-dependent properties. Folic acid, crucial for cellular growth and repair, can be paired with chitosan to create targeted nanoparticles for treating kidney stones. *Musa acuminata* stem water, rich in bioactive compounds, serves as an eco-friendly material for nanoparticle synthesis.^{1,2} Using Box-Behnken Design (BBD) allows for efficient optimization of synthesis conditions, improving the formulation and performance of folic acid-chitosan nanoparticles. This approach aims to enhance the therapeutic efficacy and stability of nanoparticles, advancing treatment strategies for nephrolithiasis.^{3,4}

MATERIALS AND METHODS

Banana stems from Assonora Bardez in North Goa were collected post-ripening. Three factors at three levels were selected and encoded for analysis. Using Box-Behnken Design, 15 experimental batches were prepared and data was collected. *M. acuminata* stems were sliced into 1 to 2 mm pieces, refluxed with distilled water for 2 hours, filtered, and the pale-yellow extract was kept at 2 to 4°C for later use.^{5,6}

Conjugation of Folic Acid with Chitosan

A 1:1 FA and triethylamine (TEA) solution in anhydrous dimethyl sulfoxide (DMSO) was combined with a 0.5% chitosan solution in 0.1 M acetic acid (pH 4.7) and stirred in

the dark at 25°C for 16 hours. After adjusting the pH to 9.0 with 1 M NaOH, the mixture was centrifuged at 2500 rpm to precipitate the FA-CS conjugate. The precipitate was dialyzed in phosphate buffer (pH 7.4) for 3 days and water for 4 days, then freeze-dried and stored.^{7,8}

Preparation of Nanoparticles (NPs)

Combining a 1:1 FA and triethylamine solution in anhydrous DMSO with a 0.5% chitosan solution in 0.1 M acetic acid (pH 4.7), the mixture was stirred at 25°C in the dark for 16 hours. After raising the pH to 9.0 with 1 M NaOH, it was centrifuged at 2500 rpm to precipitate the FA-CS conjugate. The precipitate underwent dialysis in phosphate buffer (pH 7.4) for 3 days, then water for 4 days, before being freeze-dried and stored.^{9,10}

Biosynthesis of Nanoparticles

The banana stem water extract was combined with the folic acid-chitosan conjugation solution and stirred at room temperature for 1-hour. A color change indicated nanoparticle formation.^{11,12}

Characterization of Nanoparticles

Fourier transform infrared spectroscopy (FTIR)

Using a Thermo Scientific Nicolet iS5 FTIR Spectrophotometer, the IR spectrum of the compound was recorded from 400 to 4000 cm⁻¹ to identify functional groups in the nanoparticles.

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The nanoparticles were analyzed between KBr discs to determine the chemical structure of the folic acid-chitosan conjugate.^{13,14}

Scanning electron microscopy (SEM)

A scanning electron microscope (SEM) employs a focused electron beam to produce detailed, high-resolution images of an object’s surface, revealing its composition and physical features.^{15,16}

Optimization: Box-behnken design

Three factors at three levels were selected from pre-tests and coded. The Box-Behnken model with three central points was used to randomly select and analyze 15 experimental batches.¹⁷

Optimization: Data analysis and validation

Pre-tests determined three factors at three levels, which were coded. The Box-Behnken model, including three central points, directed the selection and data collection from 15 experimental batches.¹⁸

Evaluation Studies of Nanoparticles

In vitro studies (For nephrothialiasis)

The kidney stone was weighed using a digital balance and placed in a semi-permeable cotton pouch. This pouch was suspended in a beaker containing 200 mL of filtered liquid extract, with the beaker covered and the pouch’s lower end dipped in the liquid. The setup was placed on a magnetic stirrer at 200 rpm for 2 hours at room temperature. The liquid extract was replaced every 4 days, and the weight change of the stone was recorded.^{19,20}

Antimicrobial activity of nanoparticles

• *Paper disk method*

The antimicrobial activity of nanoparticle solutions was assessed against *Staphylococcus aureus* and *Escherichia coli* using the paper disk method. Plant extract served as a negative control, and ampicillin (0.1 mg/mL) was the positive control. Inhibition zones were measured in mm after incubating the disks at 37°C for 24 hours.^{21,22}

Antioxidant activity

Table 1: Lists of independent variables

Factor	Parameter	Unit	Minimum value	Maximum value
A	Polymer	mg	300	1200
B	Sonication time	min	15	60
C	Sonication frequency	KHz	20	60

Table 2: Lists of dependent variables

Response	Name	Units
R1	Entrapment Efficiency	%
R2	Practical Yield	%

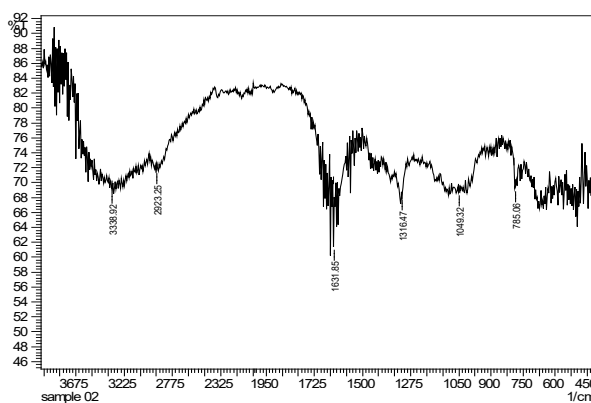


Figure 1: IR spectra of folic acid chitosan conjugate

Table 3: Interpretation of IR spectra

Functional group	Expected values	Observed value
R-CONH ₂	1690–1630 cm ⁻¹	1631.85 cm ⁻¹
O-H (ACIDIC)	3300–2500 cm ⁻¹	2923.25 cm ⁻¹
N-H (BOND STRETCH)	3600–3200 cm ⁻¹	3338.92 cm ⁻¹

Table 4: Information of nanoparticles

Compound	Morphology	Particle size
Nanoparticles	Irregular shape in cluster	80–95 nm

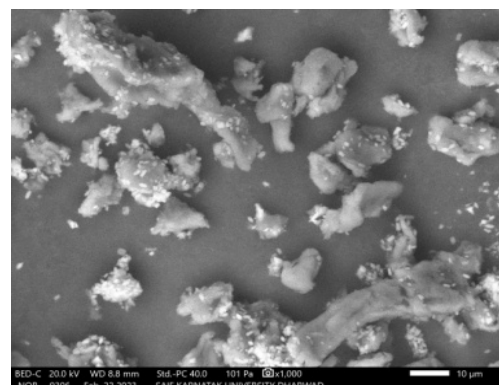
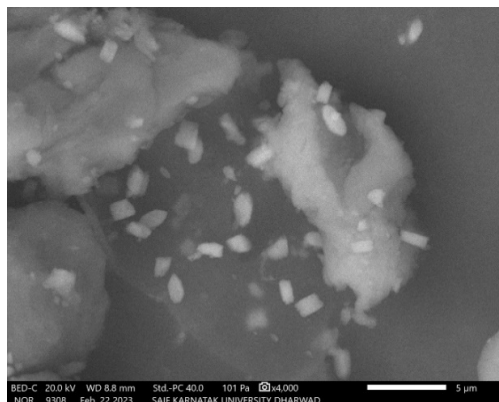


Figure 2: SEM images of nanoparticles

Table 5: Experimental Design: 15 Runs and Responses

Std	Run	Factor 1: A: Polymer (mg)	Factor 2: B: Sonication time (min)	Factor 3: C: Sonication frequency (KHz)	Response 1: Entrapment efficiency (%)	Response 2: Practical yield (%)
5	1	300	37.5	20	86.04	78.54
14	2	750	37.5	40	84.22	87.45
13	3	750	37.5	40	86.06	84.43
3	4	300	60	40	78.62	83.23
2	5	1200	15	40	89.48	78.02
12	6	750	60	60	86.34	87.93
6	7	1200	37.5	20	72.64	83.04
7	8	300	37.5	60	80.64	74.54
8	9	1200	37.5	60	82.64	89.4
15	10	750	37.5	40	86.04	84.54
1	11	300	15	40	76.76	78.54
9	12	750	15	20	83.64	89.43
4	13	1200	60	40	79.98	76.5
10	14	750	60	20	85.65	89.45
11	15	750	15	60	80.54	88.65

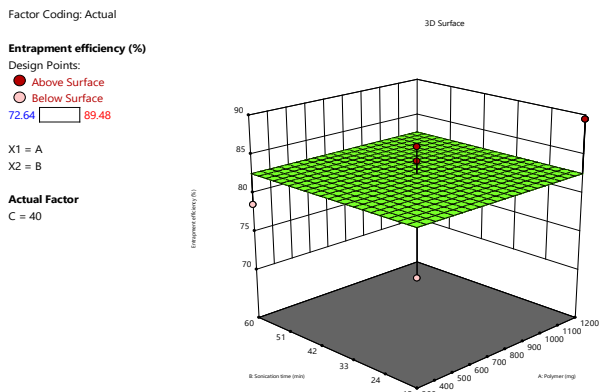


Figure 3: Counter plots of sonication time and polymer versus entrapment efficiency

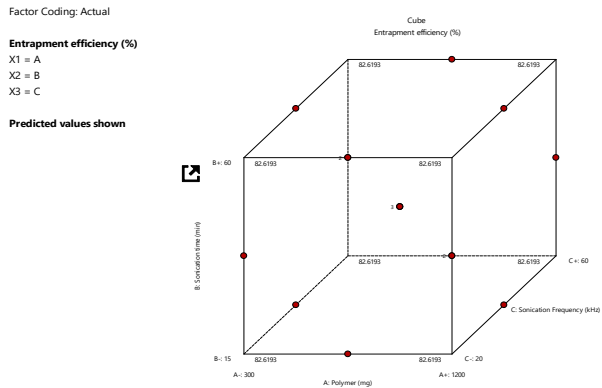


Figure 4: Cubic design illustrating the interaction between polymer, sonication time, and frequency on entrapment efficiency

Table 6: ANOVA analysis for entrapment efficiency

Source	Sum of squares	Df	Mean square	F-value	p-value
Model	0.0000	0			
Residual	274.06	14	19.58		
Lack of Fit	271.83	12	22.65	20.29	0.0479 Significant
Pure error	2.23	2	1.12		
Cor total	274.06	14			

• *Hydrogen peroxide scavenging*

A 0.1 M H₂O₂ solution in phosphate buffer (pH 7.4) was prepared for the hydrogen peroxide scavenging assay. The extract was combined with 0.6 mL of this H₂O₂ solution in 3.4 mL of phosphate buffer, and the absorbance was measured at 230 nm.^{23,24}

RESULTS AND DISCUSSION

Characterisation

FTIR analysis

Results are depicted in Figures 1 and Table 3.

Sem analysis

SEM analysis revealed that the nanoparticles have an irregular, clustered shape with a size range of 80 to 95 nm.

Optimization process variables of nanoparticle

Results are depicted in Table 5.

Table 7: ANOVA analysis for practical yield

Source	Sum of squares	Df	Mean square	F-value	p-value	
Model	211.07	2	105.54	7.60	0.0074	Significant
A-Polymer	18.33	1	18.33	1.32	0.2731	
A ²	192.74	1	192.74	13.87	0.0029	
Residual	166.71	12	13.89			
Lack of fit	160.84	10	16.08	5.48	0.1640	Not significant
Pure error	5.87	2	2.93			
Cor total	377.78	14				

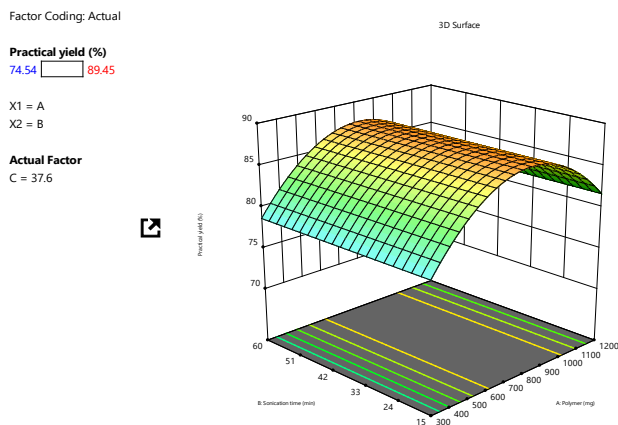


Figure 5: Counter plots of sonication time and polymer versus practical yield

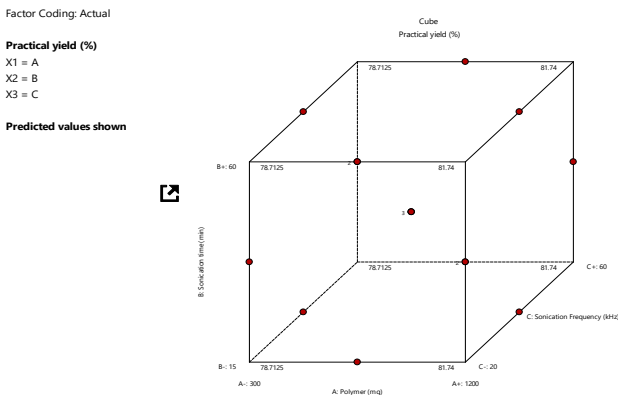


Figure 6: Impact of polymer type, sonication time, and frequency on entrapment efficiency in cubic design

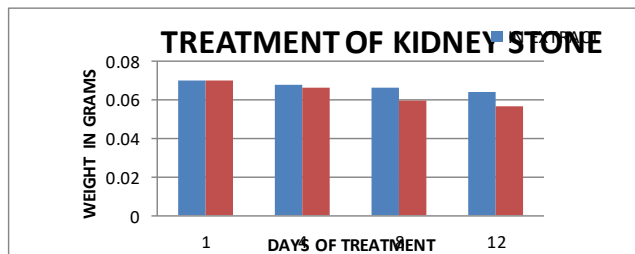


Figure 7: Comparison of stone weight in extract and nanoparticles

Anova for Mean Model

Response 1: Entrapment efficiency

Results are depicted in Table 6.

Factor coding is Actual

Sum of squares is type III - partial

In this case, no model terms are significant, as *p-value* are all above 0.0500. Terms with *p-value* above 0.1000 are deemed insignificant and may necessitate model reduction. Additionally, a Lack of Fit F-value of 20.29, with a 4.79% chance of occurring randomly, highlights a significant lack of fit, which is not ideal for achieving a good model.

ANOVA for Reduced Quadratic mode

Response 2: Practical yield

Results are depicted in Table 7.

Factor Coding is Coded

Sum of squares is type III - Partial

With a Model F-value of 7.60, there is only a 0.74% chance of obtaining such a large value by chance, indicating the model is significant. Significant model terms, including A², are identified by *p-value* below 0.0500, while terms with values above 0.1000 are not significant and may require model reduction. The Lack of Fit F-value of 5.48, with a 16.40% chance of occurring by chance, is not significant relative to pure error, suggesting the model fits well.

By performing optimization, 9 solutions were obtained out of which one was used for preparation.

Optimization, Validation and Evaluation

Design Expert recorded and analyzed optimized formulations (Table 9) using a mathematical model. The formulation with the best process conditions and desirability factors was selected. Under these optimized conditions, the solutions and nanoparticles were prepared. The prepared nanoparticles were compared with the plant extract. SEM analysis was performed to assess particle size and morphology. *In-vitro* studies for nephrolithiasis and comparative evaluations between nanoparticles and plant extract were conducted.

Evaluation of *in-vitro* studies (for nephrothlialiasis)

Results are depicted in Table 9 and Figure 7.

Antimicrobial Activity

No zone of inhibition was observed for plant extract or nanoparticles against *E. coli* and *S. aureus*.

Only the positive control exhibited antimicrobial activity; neither *E. coli* nor *S. aureus* showed any inhibition zones for the nanoparticles or plant extract.

Antioxidant Activity

Antioxidant activity of the extract and nanoparticles was assessed *via* H₂O₂ scavenging at 230 nm.

From the above result, nanoparticles show more antioxidant activity than the extract.

Table 8: Solutions for optimum formulations

Number	Polymer	Sonication time	Sonication frequency	Entrapment efficiency	Practical yield	Desirability	Remark
1	300.000	37.500	40.000	82.619	77.017	1.000	Selected
2	750.000	37.500	40.000	82.619	85.473	1.000	
3	1200.000	37.500	40.000	82.619	80.044	1.000	
4	759.239	37.500	40.000	82.619	85.501	1.000	
5	456.636	37.500	40.000	82.619	81.536	1.000	
6	874.474	37.500	40.000	82.619	85.361	1.000	
7	736.279	37.500	40.000	82.619	85.421	1.000	
8	1173.314	37.500	40.000	82.619	80.753	1.000	
9	1084.016	37.500	40.000	82.619	82.772	1.000	

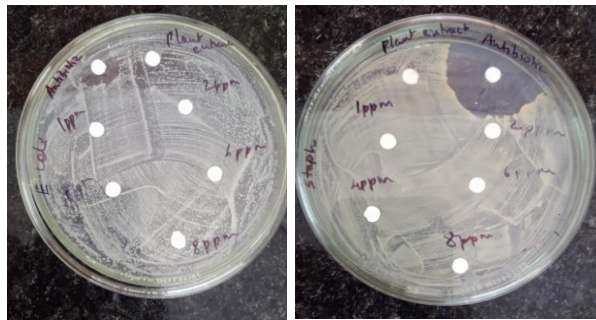


Figure 8: Inoculated agar plates of *E. coli* and *S. aureus*

Table 9: Comparison of stone weight in extract and nanoparticles

Days of treatment	Stone weight in grams	
	In extract	In nanoparticles
1	0.070	0.070
4	0.068	0.066
8	0.066	0.060
12	0.064	0.057

Table 10: Antimicrobial activities

Culture	Zone of inhibition (nm)	
	Extract	Nanoparticles
<i>E. coli</i>	-	-
<i>S. aureus</i>	-	-

Table 11: Antioxidant activity of extract

Tube	Absorbance At 230 nm	% of antioxidant
Control	0.972	-
Extract	0.550	43.415

Table 12: Antioxidant Activity of Nanoparticle

Tube	Absorbance At 230 nm	% of antioxidant
Control	1.092	-
Nanoparticle	0.461	57.783

SUMMARY

Addressing kidney stone recurrence, this study developed nanoparticles by conjugating folic acid with chitosan and reacting them with *M. acuminata* stem extract. FTIR confirmed amide bond formation, while SEM revealed irregular, clustered nanoparticles of 80 to 100 nm. Optimization via Box-Behnken Design improved production efficiency, resulting in nanoparticles with a mean diameter of 80 to 90 nm. In vitro tests showed these nanoparticles were more effective in dissolving kidney stones than the plant extract. Both showed no antimicrobial activity, but the nanoparticles had superior antioxidant activity

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