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Research Article

Spectrophotometric Estimation of Torsemide in Tablet Dosage Form Using Chemical Derivatization Technique

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ABSTRACT

A simple and cost effective spectrophotometric method is described for the determination of Torsemidein pure form and in pharmaceutical formulations. The method is based on the formation derivative when the drug reacts with picric acid solution. The colored derivative solution has an absorption maximum at 350 nm; also obeys Beer's law in the concentration range $5-25\mu g$ /mL. The absorbance was found to increase linearly with increasing concentration of torsemide, which is corroborated by the calculated correlation coefficient value of 0.997. The slope and intercept of the equation of the regression line are 0.035 and 0.002 respectively. The optimum experimental parameters for the reaction have been studied. The validity of the described procedure was assessed. Statistical analysis of the results has been carried out revealing high accuracy and good precision. The proposed method was successfully applied to the determination of Torsemide in pharmaceutical formulations.

Key words:

Torsemide, chemical derivatization, UV-Spectrophotometry

INTRODUCTION

Torsemide¹ is a loop diuretic and chemically known as 3-pyridine sulfonamide N-[[(1-methylethyl) amino] carbonyl]-4-[(3-methylphenyl) amino]. It acts by inhibiting the Na $^+$ / K $^+$ / 2Cl $^-$ carrier system in the lumen of the thick ascending portion of the loop of Henle, resulting in the decrease in reabsorption of sodium and chloride. It is commonly used in cases of congestive heart failure and in kidney disease. Literature survey reveals that, few Spectroscopic methods² for estimation in formulation and chromatographic methods³, have been reported for the estimation of Torsemide in human plasma and urine⁴. To the best of our knowledge, there is no spectrophotometric in the literature reported using chemical derivatisation technique for torsemide. The objective of the present is to develop simple and sensitive work spectrophotometric method for the estimation of Torsemidein pure drug and in pharmaceutical formulations.

From Below graph it can be seen that Torsemide drug showed less absorbance at 284nm (λ_{max}) so derivatisation of Torsemide was done to improve absorbance of Torsemide.(Fig. No.01)

The method was validated as per standard ICH guidelines. ^(5, 6)

EXPERIMENTAL⁷⁻¹⁴

Instruments and Chemicals

UV-Visible Spectrophotometer (Perkin-Elmer, Lambda 25) was used for spectral measurements with spectral

*Corresponding Author Email: <u>sbbagade@rediffmail.com</u> band width 1 nm; wavelength accuracy is 0.5 nm and 1 cm matched quartz cells.

All chemicals used were of analytical reagent grade and methanol was used throughout. Dytor and Tide are then commercial tablet formulations labeled to contain 10 mg of torsemide per tablet. Stock reference solution was freshly prepared from pure sample of torsemide derivative by dissolving 25mg in 25 mL of methanol.

 Table1: Regression and Optical characteristics of derivative of Torsemide

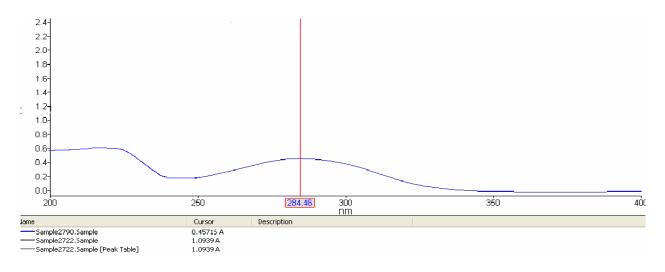
Parameters	Value
Selected analytical wavelength	350 nm
Beer's law range	5-25 μg mL ⁻¹
Coefficient of correlation	0.999
Slope	0.035
Intercept	0.002

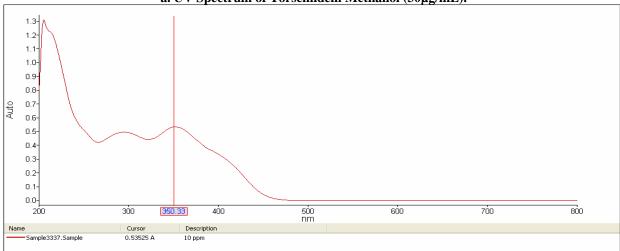
Preparation of Picric Acid Derivative of Torsemide 0.5 mg of pure drug was dissolved in 5ml of hydrochloric acid and added about 5 ml of saturated solution of picric acid in water. This above mixture was then warmed for 5 minutes on the water bath and then allowed to cool to room temperature, for crystallization. The precipitate obtained above was then subjected to recrystalisation by using water or methanol as solvent. The yellow crystals obtained were

of ionic bond of picric acid with the drug. (Fig. No. 2) **Preparation of solutions**

Accurately weighed 25 mg of Torsemide derivative transferred to 25mL volumetric flasks and the resultant solution was further diluted to get100ppm solution.

In to 10 mL measuring flasks, different aliquots of working standard solution (1.0- 5.0 mL) were transferred to provide final concentration range $5.0 - 25.0 \mu/g$ mL.





a. UV Spectrum of Torsemidein Methanol (30µg/mL).

b. UV Spectrum of derivative of Torsemidein methanol. (15µg/mL) Fig.1: Difference between UV specrum of Torsemide derivative (b) and Torsemide (a).

Table 2: Results of analysis of tablet by dramatization				
Pharmaceutical formulation	% Label Claim in mg	%Average	S.D.	%RSD
Tablet (Dytor)	10	99.22%	0.389	0.392
Tablet (Tide)	10	100.03%	0.209	0.209

The solutions were made up to volume with distilled water. The absorbance of each solution was measured at 350 nm against the reagent blank. The calibration graph was then prepared by plotting the absorbance versus concentration of the drug. (Fig. No.3)

Table 3: Results of Recovery Study			
% Solution	% Recovery± SD	%RSD	
For 80 %	$99.16 \pm 0.588\%$	0.590	
For 100 %	99.87 ±1.59%	1.592	
For 120 %	$101.19 \pm 0.392\%$	0.387	
Ruggedness	101.07 ±0.301 %	0.297	
Robustness	97.77 ±0.224%	0.224	

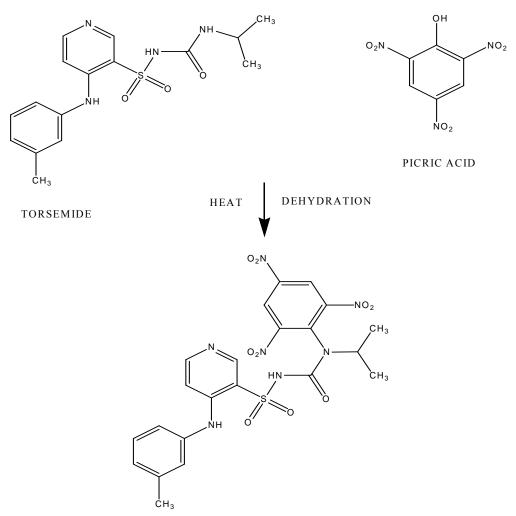
Assay of tablet formulation

In order to see the feasibility of proposed method for estimation of Torsemide in marketed pharmaceutical formulations, for analysis of commercial formulation; twenty tablets were weighed, average weight determined and crushed into fine powder. A quantity of tablet powder equivalent to 10 mg of Torsemide was transferred into 100 ml glass volumetric flask containing 50 ml water, shaken manually for 10 min, volume was adjusted to mark with same solvent and filtered through whatmann filter paper no.1. The appropriate aliquots were transferred to 10 ml glass volumetric flask; volume was adjusted to the mark with same solvent to obtain concentration of 15 μ g/mL. The absorbance of the solutions were recorded at 350nm and the concentration of the torsemide was determined. % label claim was then calculated; results are shown in Table No. 1.

Table 4: Results of Precision			
Precision (as % Relative standard deviation)			
Parameter	Avg±SD	%RSD	
Interday Precision	0.1881±0.185 %	0.188	
Intraday Precision	0.2173±0.213%	0.217	
Repeatability	0.5647±0.558%	0.567	
Validation		0.007	

Validation

The Proposed method was validated as per the USP



PICRATE DERIVATIVE OF TORSEMIDE Figure 2: Formation of ionic bond with picric acid

guidelines for linearity and accuracy, precision **Recovery studies**

To study the accuracy of the proposed methods and to check the interference from excipients present in the dosage form, recovery experiments were carried out by the standard addition method. Recovery studies were carried out by standard addition method at three different levels. A known amount of drug was added to preanalyzed capsule powder and percentage recoveries were calculated. The results of recovery studies were satisfactory and are presented in Table No.3

Table 5: Results of Specificity			
Condition (24hrs)	Wt. taken In mg	% rug estimated	% Degradati on
0.1N HCl	25.16	61.48	38.52
0.1N NaOH	24.80	37.70	62.30
$3\% H_2O_2$	25.62	47.77	52.23
Heat(60 ⁰ C)	25.41	59.92	40.08
Nomal	25.09	97.43	96.56

Precision

Precision of the method is studied as repeatability, intraday and inter-day precision. Repeatability was determined by analyzing Torsemide derivative (15µg/mL). For six times. Intra-day precision was determined by analyzing the 15 μ g/mL of Torsemide derivative for three times in the same day. Inter-day precision was determined by analyzing the same concentration of the solutions daily for three days, results are reported in Table No.4.

Table No. 6: Results of Ruggedness and Robustness		
Ruggedness	101.07 ±0.301 %	0.297
Robustness	$97.77 \pm 0.224\%$	0.224

Specificity

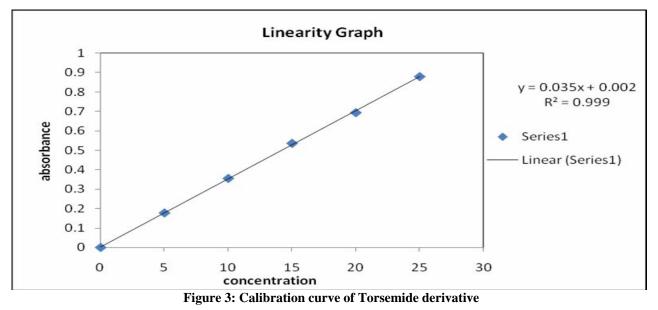
The specificity study was performed for the proposed analytical method by the preparation of degradants such as 0.1N HCL, 0.1N NaOH, 3% H₂O₂, Heat and UV Light by subjecting the analyte to different degradation conditions for 24 hours and results were recorded and are depicted in Table No 5.

Ruggedness

The studies of ruggedness were carried out by using two different analysts.

a) Different analyst

The sample solutions were prepared by two different analysts and same procedure was followed as described earlier. The % label claim was calculated as done in marketed formulation estimation. Results and statistical data are shown in Table No.6. **Robustness**



The robustness of an analytical procedure is studied by a measure of its capacity to remain unaffected by small, but deliberate variations in method parameters. Robustness was studied by taking absorbance of 15 mcg/mL solution at lower wavelength for five times. Results and statistical data are shown in Table No.6.

RESULTS AND DISCUSSION

The method developed for spectrophotometric determination of Torsemide in tablet formulation was found to be simple and convenient for the routine analysis. Beer-Lambert law was obeyed in the concentration range of 5-25 mcg/ml. Co-efficient of variation was found to be 0.997. The percentage recoveries were found in the range of 100.25%. While the method was found to be precise with % relative standard deviation \pm (0.1881 %) for interday precision and for intraday (0.2173%).

CONCLUSION

The proposed method is simple, rapid, precise and inexpensive, and hence can be used in routine analysis of Torsemide in bulk drug and in formulations. It can be easily conveniently applied for quality control analysis.

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REFERENCES

 Blose, J. L., Adams. K. F. and Patterson, J. H. 1995. Torsemide: a pyridine-sulfonylurea loop diuretic . Ann Pharmaco Ther. 29:396-402.

- Krishna, M. V., Sankar. D.G. 2007. Simple spectrophotometric determination of torsemidein bulk drug and in formulation. European Journal of Chemistry.5:73-478.
- Barroso, M.B., Alonso R.M., Jiménez and R. M. 2001. Simultaneous Determination of Torasemide and Its Major Metabolite M5 in Human Urine by High-Performance Liquid Chromatography–Electrochemical Detection. Journal of Chromatographic Sciences, 6:491-496.
- Engelhardt, S., Meineke. I. and Brockmöller. J. 2004. Improved solid-phase extraction and HPLC measurement of torasemide and its important metabolites. Journal of Chromatography B. 831: 31-35.
- ICH –Guidelines Q2A, Validation of Analytical Procedures: Definition and terminology (CPMP III/5626/94) March (1995) Geneva, Switzerland.
- ICH –Guidelines Q2B, Validation of Analytical Procedures: Methodology (CPMP/ICH/281/95) November (1996) Geneva, Switzerland.
- Maryadele, J. O'Neil, Patrica, Heckelman, E. Koch, C. B., Raman, K. J., Kenny, C. M. and D'Arecca R. 2006. The Merck Index An Encyclopedia of Chemicals, Drugs and Biologicals, 14th ed, Vol-1, 275. UK Longman Group Ltd.
- United State Pharmacopoeia National Formulary, published by United State Convention 2009, vol-3 pp-3774
- Willard, H. H., Merritt, L. L., Dean, S. A. and Settle, F. A., 2006. Instrumental Methods of Analysis; 7th Edition, India. CBS Publishers & Distributors. 148.
- Skoog, West and Holler 1992. Fundamental of Analytical Chemistry, 7thed, Australia. Saunders College Publishing, 147.
- Chatwal, G. R. and Anand, S. K. 2004. Instrumental Methods of Chemical Analysis, 5thed, India. Himalaya Publishing House. 2.107-2.110.
- Beckett, A. H. and Stenlake, J. B. 1997. Practical Pharmaceutical Chemistry, 7th ed, Part Two, India. CBS Publishers & Distributors. 75.
- Furniss, B., Hannaford, A. J., Smith, P. W. and Totchell, A. R. 2007. Vogel's Textbook of Practical Organic Chemistry, 5thed, Pearson Education, 835, 929-927.
- Chan, C. C., Lam, H. and Lee Y. C. 2004. Analytical method validation and instrument performance verification, A John wiley & sons, inc., publication, 11-15